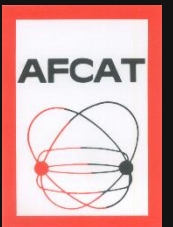




CHARACTERIZATION OF OPAQUE AND
CONCENTRATED EMULSIONS: A
CHALLENGE FOR LIQUID/LIQUID
SEPARATION

CARACTÉRISATION DES ÉMULSIONS
OPAQUES ET CONCENTRÉES: UN
CHALLENGE POUR LA SÉPARATION
LIQUIDE/LIQUIDE

CHRISTINE DALMAZZONE – IFPEN



JCAT53, PALAISEAU – 22-24 MAI 2023

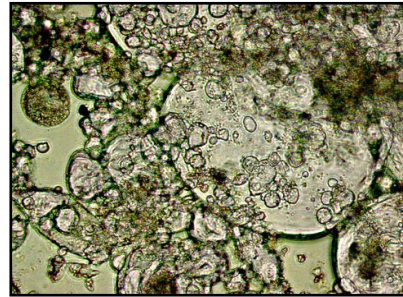
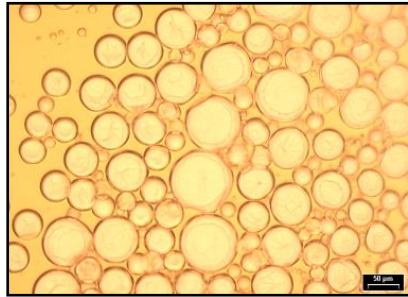


CONTEXT

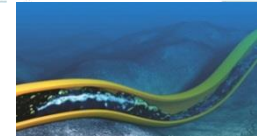
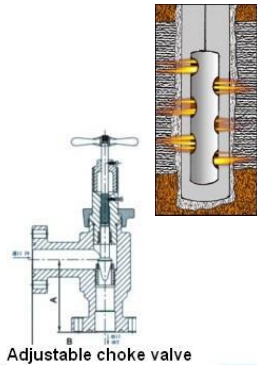
- Liquid/Liquid separation processes
 - Oil dehydration
 - Deoiling of waste waters
 - Breaking of « chocolate mousse » after an oil spill
 -



LIQUID/LIQUID SEPARATION: EMULSION ISSUES



DURING PRODUCTION, EMULSIONS ARE FORMED:



Through perforations
(comingled oil/water/gas)

Through choke valves

By turbulence (multiphase lines)

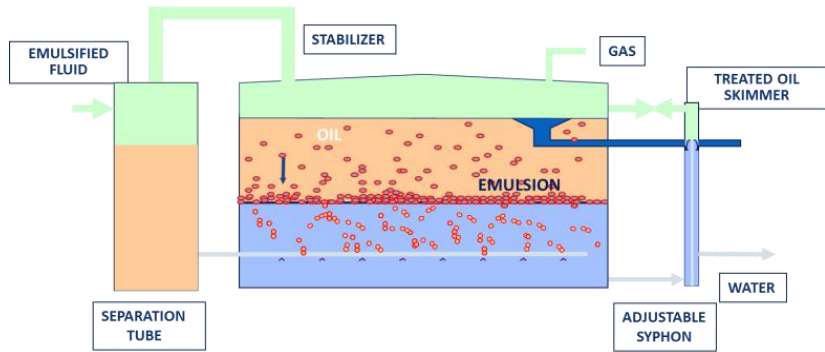
With some chemicals (surfactants, emulsifying agents)
or solid particles (sand, silt, clay....)

Everywhere when
a sufficient amount of
energy is transferred
to the system

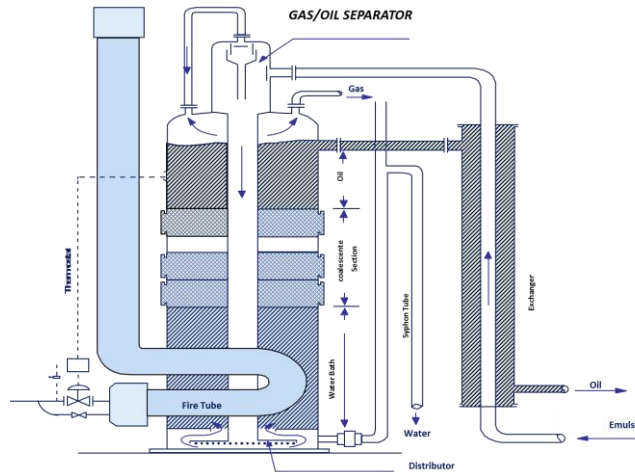


Study of droplet
breakup through a
calibrated orifice

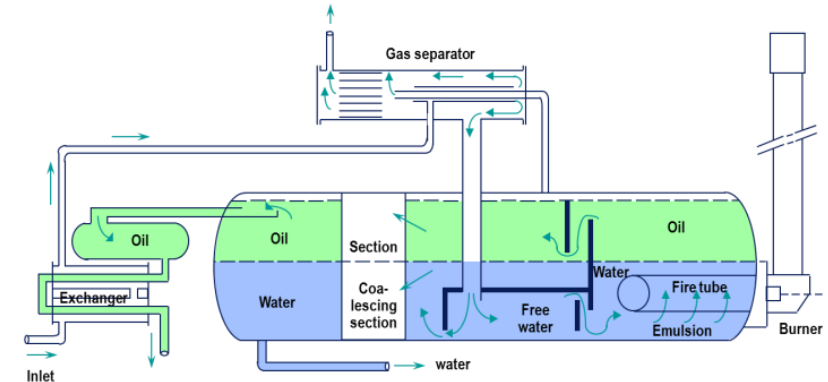
OIL DEHYDRATION: CLASSICAL TREATMENTS



Wash Tank

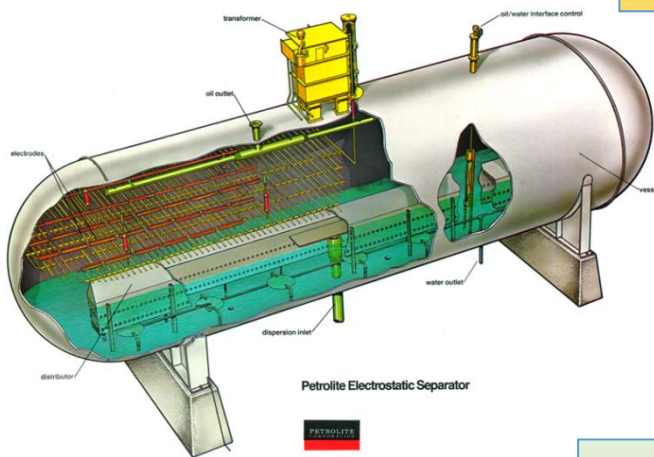


Vertical Heater Treater

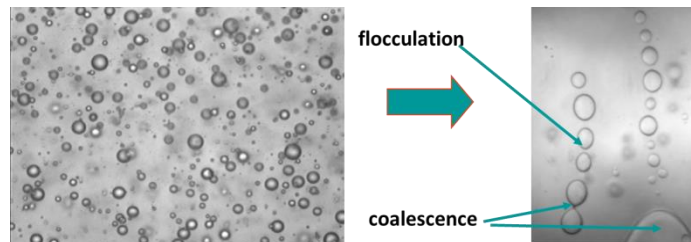


Horizontal Heater Treater

Gravitational Separation / Heating / Electrocoalescence / Demulsifiers



Electrostatic dehydrator or desalter



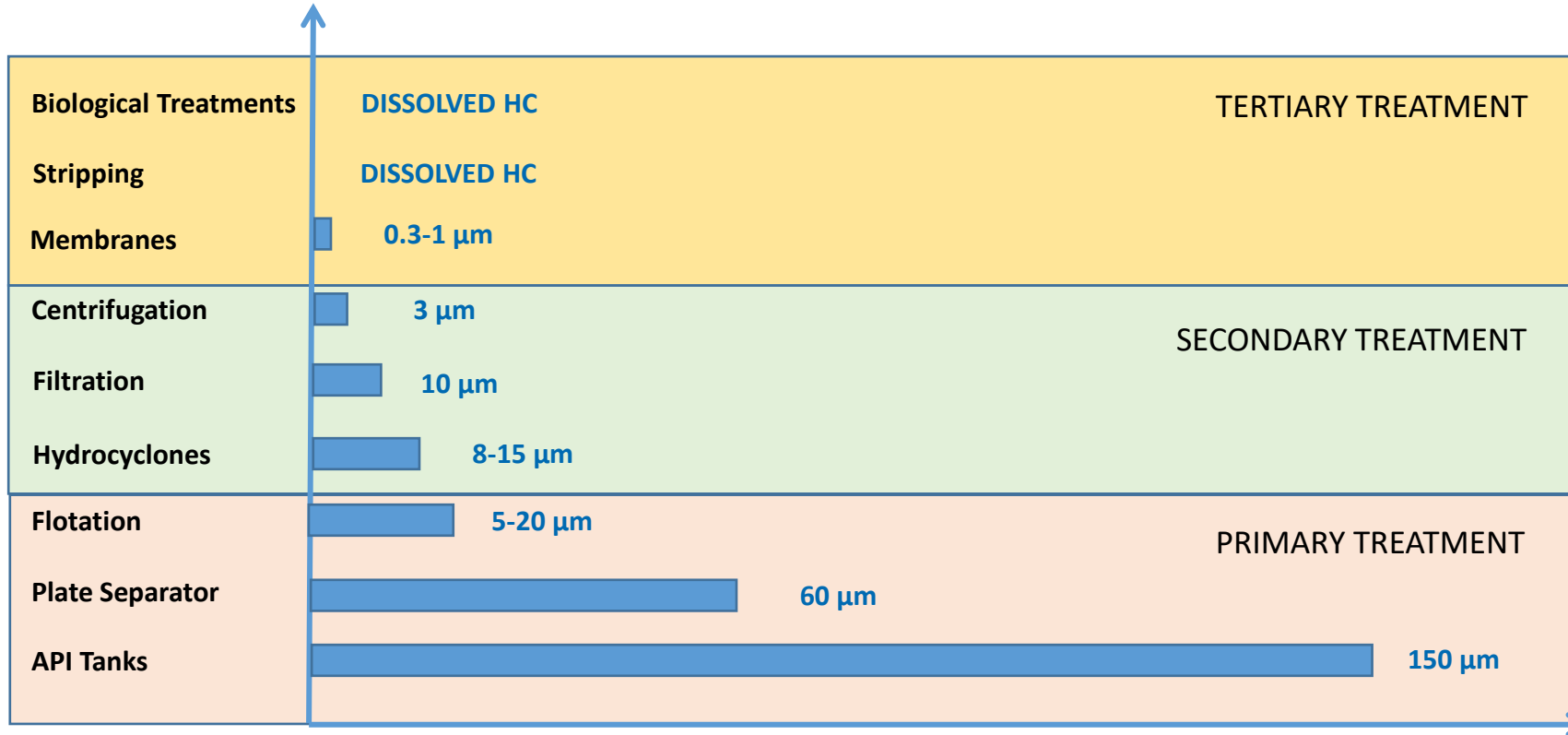
flocculation

coalescence

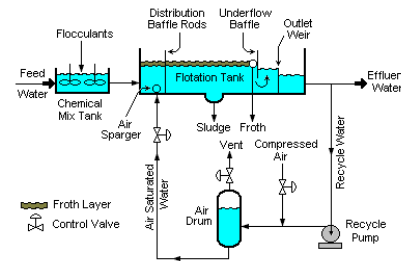
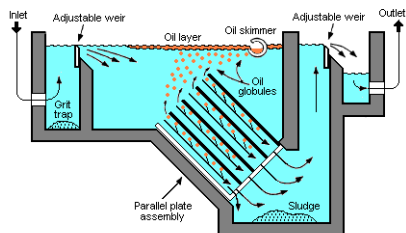


Chemical Demulsifier

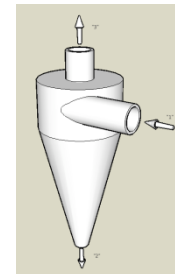
WATER DEOILING: CLASSICAL TREATMENTS



Source: Alfa Laval

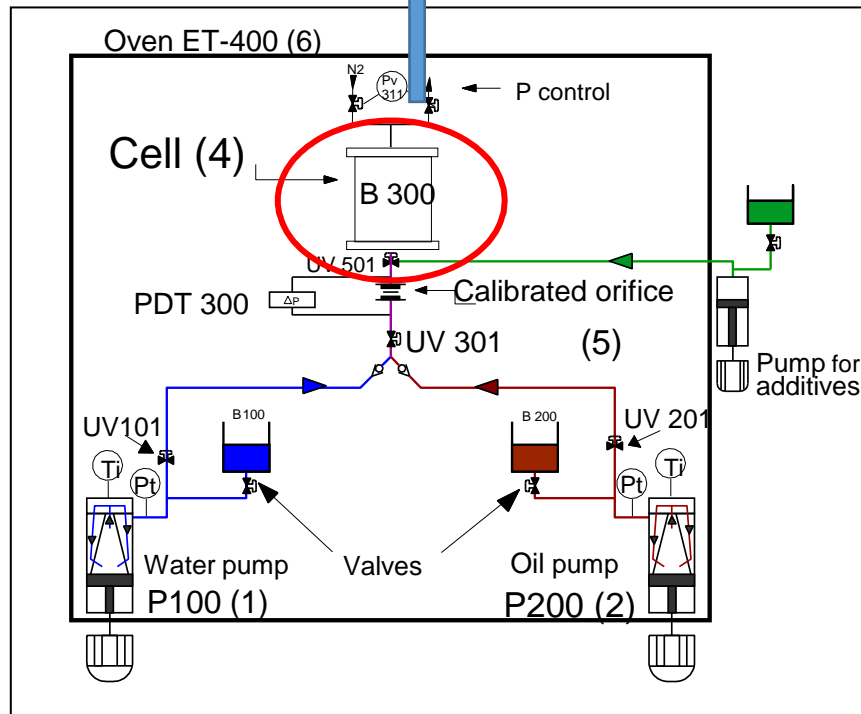


Source: Wikipedia



OIL PARTICLE DIAMETER (μm)

GENERATION OF REPRESENTATIVE EMULSIONS

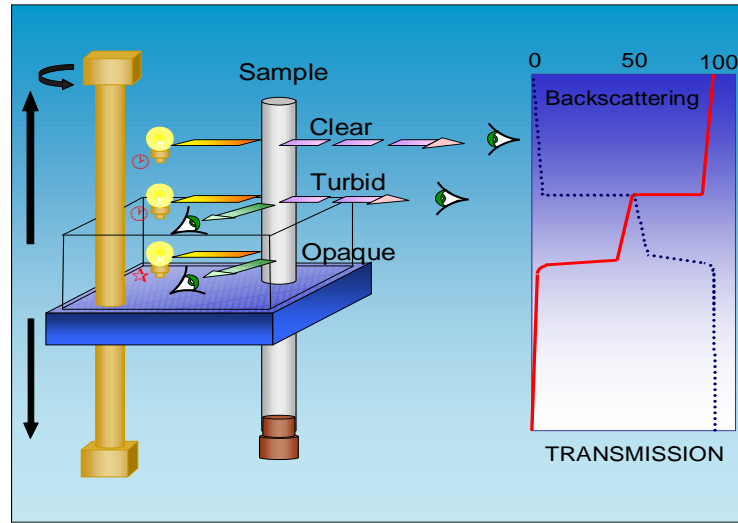
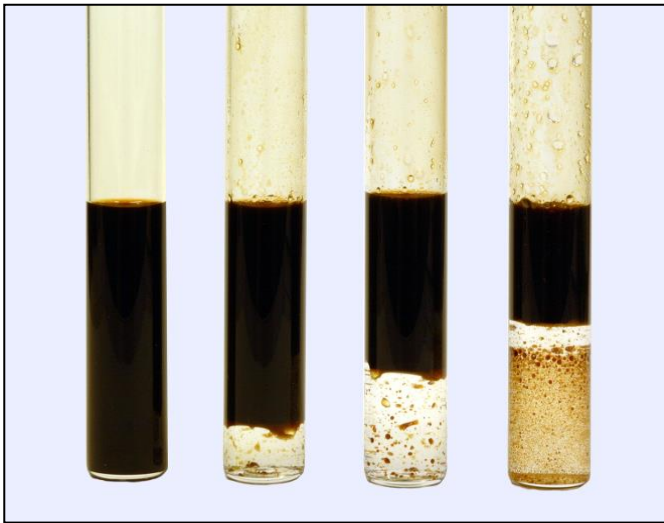


IFPEN DISPERSION RIG WITH INSTRUMENTED SEPARATOR

Janssen, Noik and Dalmazzone (2001), SPE n°71473

CHARACTERIZATION OF EMULSIONS - STABILITY

CLASSICAL BOTTLE TESTS IN THE LAB



Source: Formulation

Particularly suitable for the study of concentrated and opaque crude oil emulsions



Kinetics of emulsion separation

Mode of action of demulsifiers

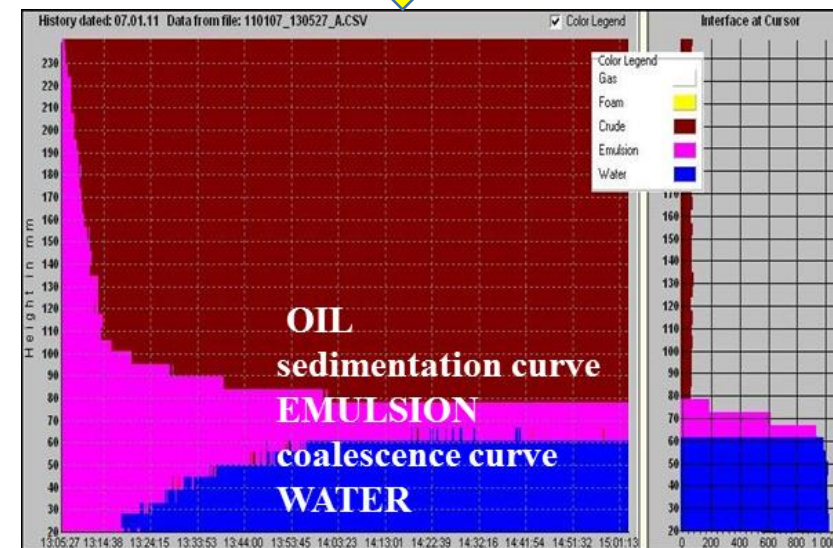
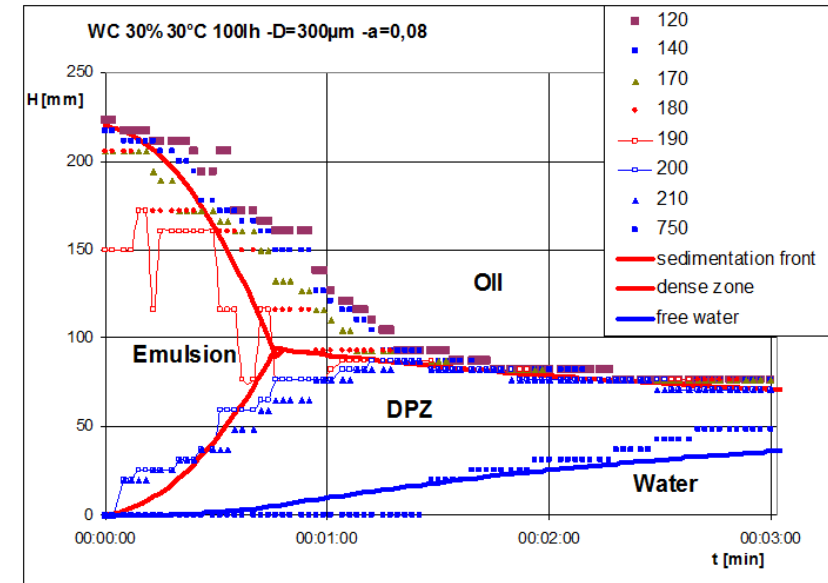
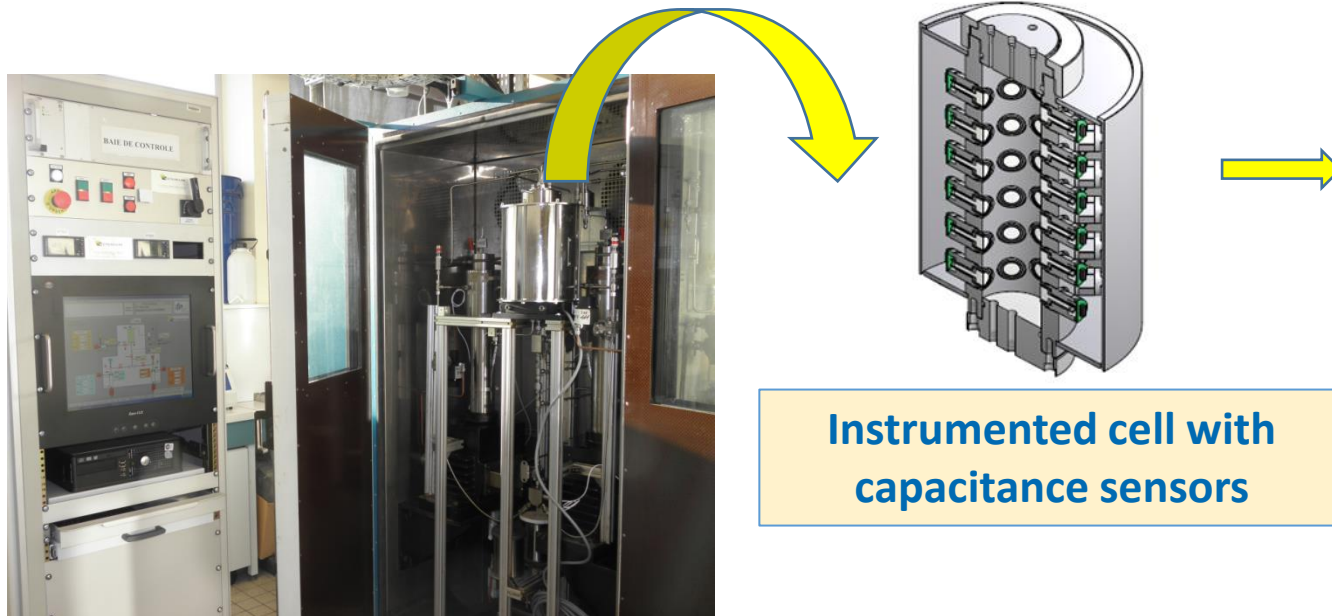
Clarity of separated water

Thickness of the o/w interface

Monitoring of the evolution of phase separation by using transmitted and backscattered light (Turbiscan™)

Dalmazzone and Noik (2001), Development of New "green" Demulsifiers for Oil Production, SPE n°65041

CHARACTERIZATION OF EMULSIONS - STABILITY



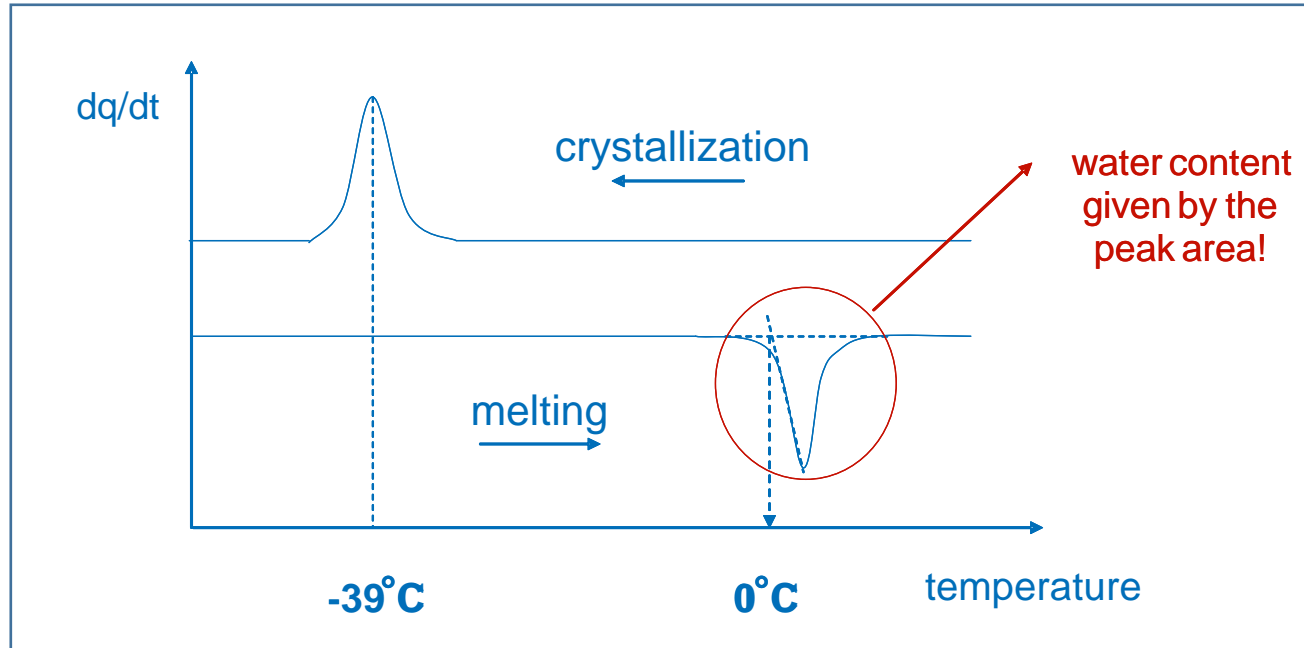
Noïk, Palermo and Dalmazzone (2013), *Journal of Dispersion Science and Technology* **34**, 1029.

CHARACTERIZATION OF EMULSIONS – STABILITY - DSC

Differential Scanning Calorimetry (DSC) may be advantageously used to characterize complex water-in-oil (w/o) emulsions.

Peaks of crystallization: the lower the temperature of ice crystallization, the smaller the droplets

Peaks of melting: the peak area gives the water content



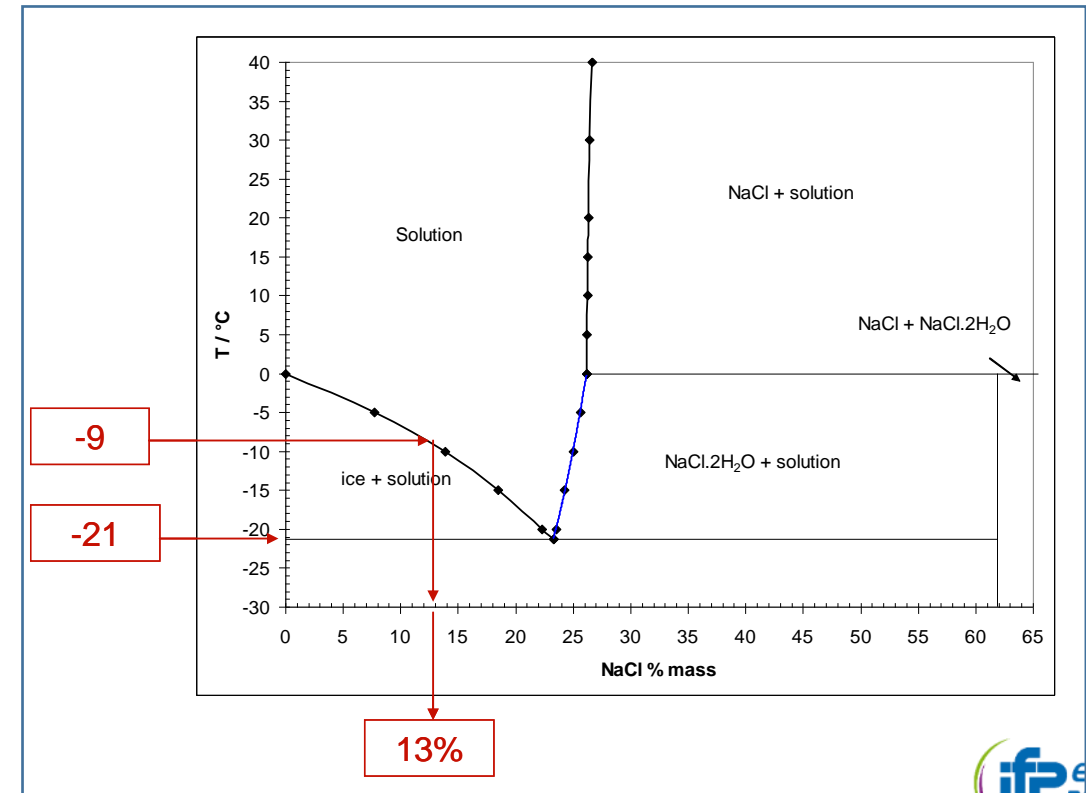
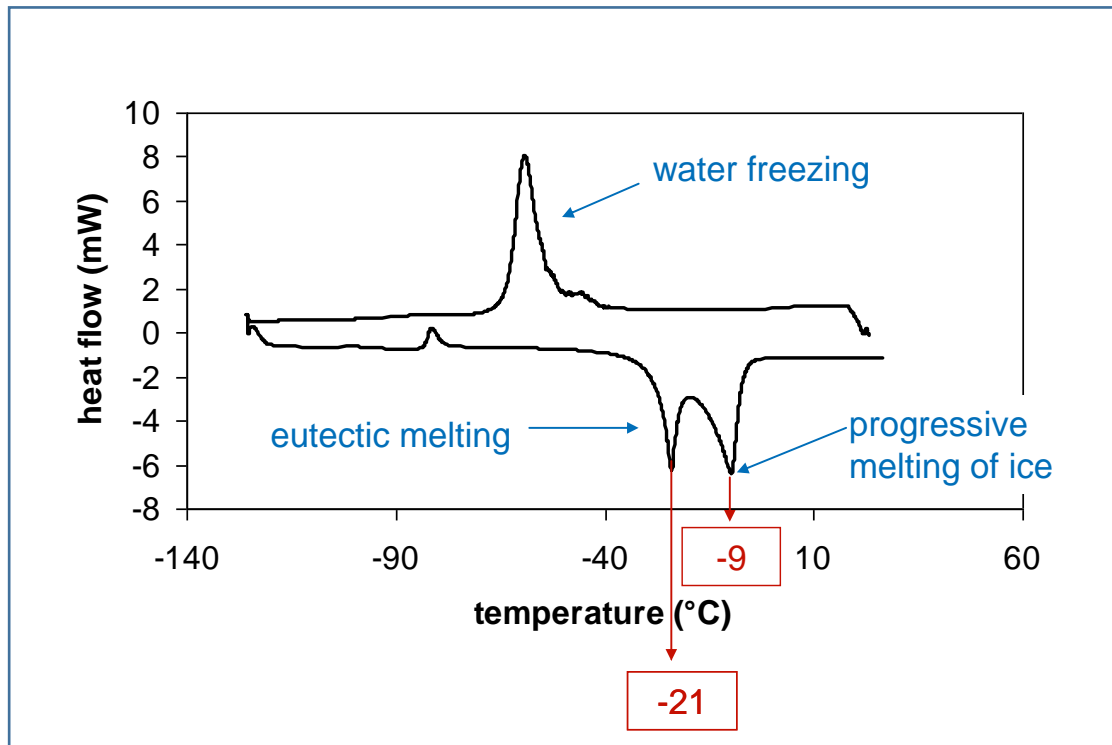
Dalmazzone, Noik and Clause (2009),
Application of DSC for Emulsified System
Characterization, *Oil & Gas Science and
Technology* **64**(5), 543.

CHARACTERIZATION OF EMULSIONS – STABILITY - DSC

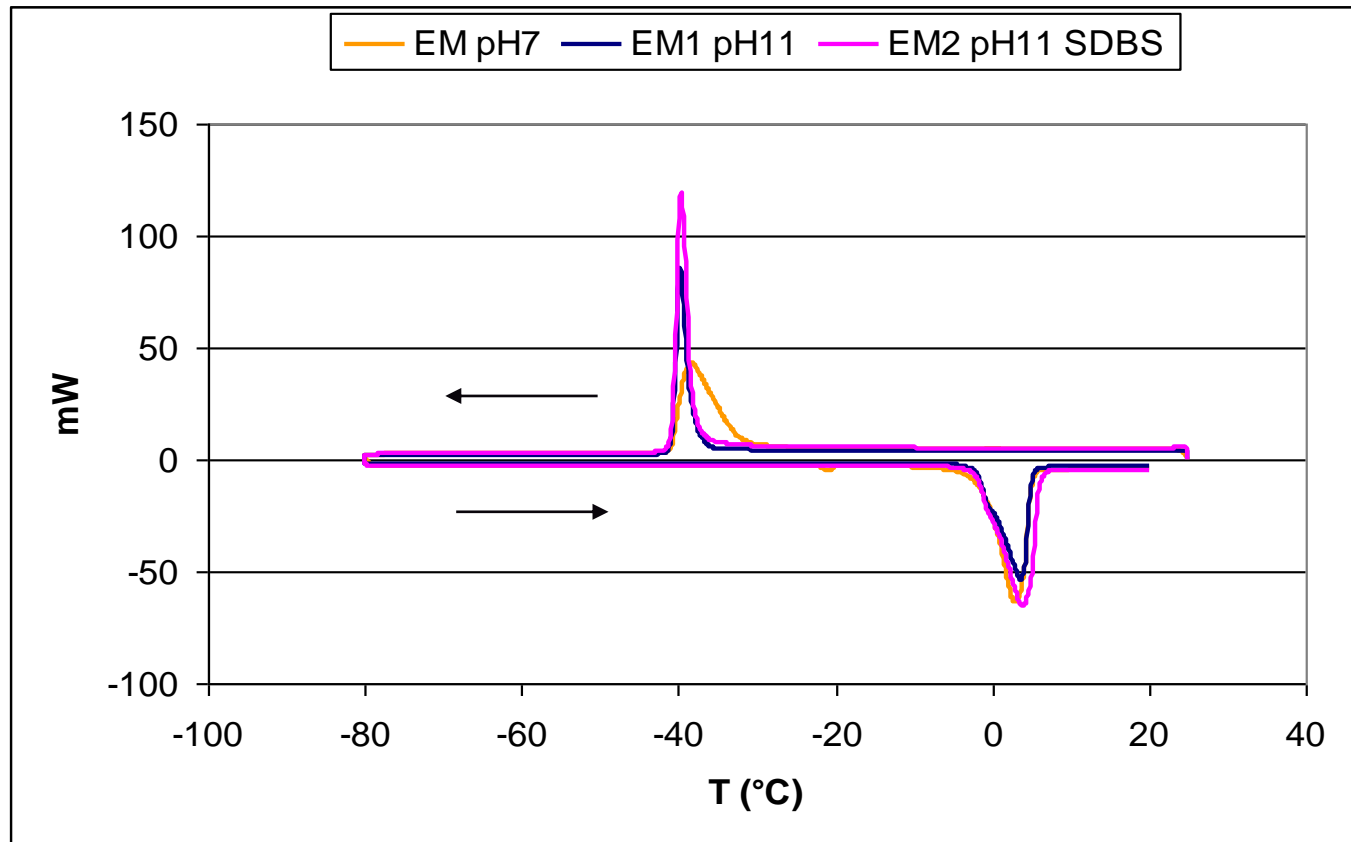
In the case of brine-in-oil emulsions:

the temperature of end of ice melting gives the composition of the aqueous phase for a binary mixture water+solute (salt for example)

for salts like NaCl, the binary phase diagram may be used to determine the composition from DSC thermograms



CHARACTERIZATION OF EMULSIONS – STABILITY - DSC



EM: Emulsion pH7,
5 g/L NaCl

EM1: Emulsion pH11,
10 g/L Na₂CO₃

EM2: Emulsion pH11,
10 g/L Na₂CO₃,
0.1% SDBS (sodium
dodecylbenzene sulfonate)

**Comparison of Water-in-Heavy Oil
Emulsions**

Dalmazzone *et al.* (2012), *Energy & Fuels*
26, 3462.

CHARACTERIZATION OF EMULSIONS – DROPLETS SIZE

Encapsulation of the droplets in order to avoid coalescence

Interfacial polymerization

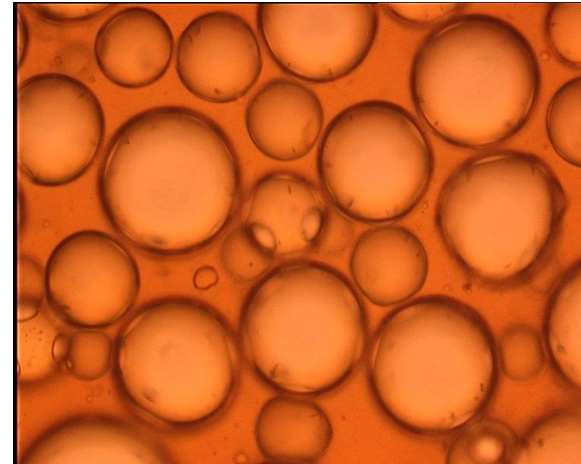
hydrophilic reagent: triethylene tetramine

hydrophobic reagent: terephthaloyl dichloride

Optical Microscopy

+

Image Analysis



Noïk *et al.* (2004), *Oil & Gas Science and Technology* **59**(5), 535.

CHARACTERIZATION OF EMULSIONS – DROPLETS SIZE

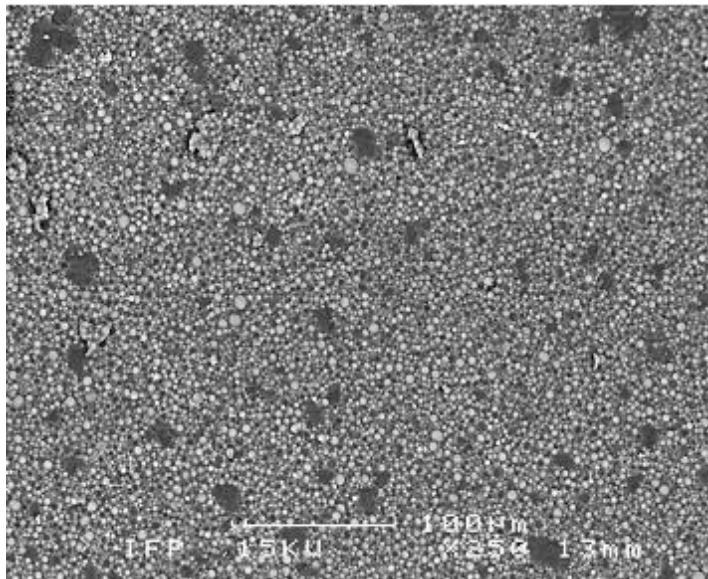
CRYO-SEM: SEM fitted with a cold stage unit

Freezing of emulsion in a nitrogen slush at -200°C

Transfer in a cryo lock chamber at -170°C

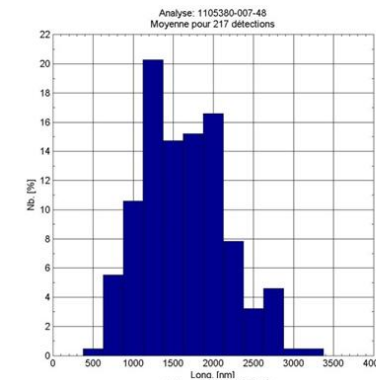
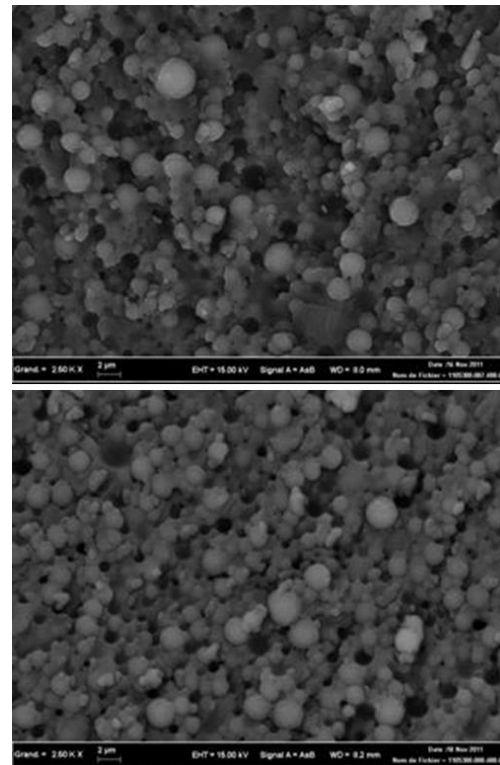
Fracturation and metallization with Cr

Transfer to the microscope chamber at -180°C

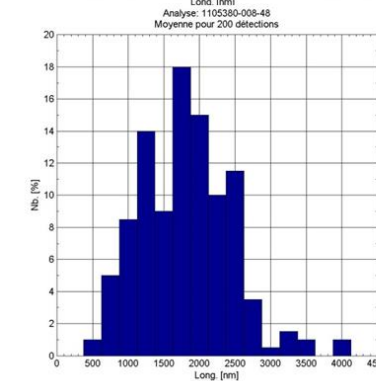


Determination of droplet size distribution in the emulsion

Image analysis software



EM1
Average Diameter
1.6 μm



EM2
Average Diameter
1.8 μm

Limit of Detection: 400 nm!!!

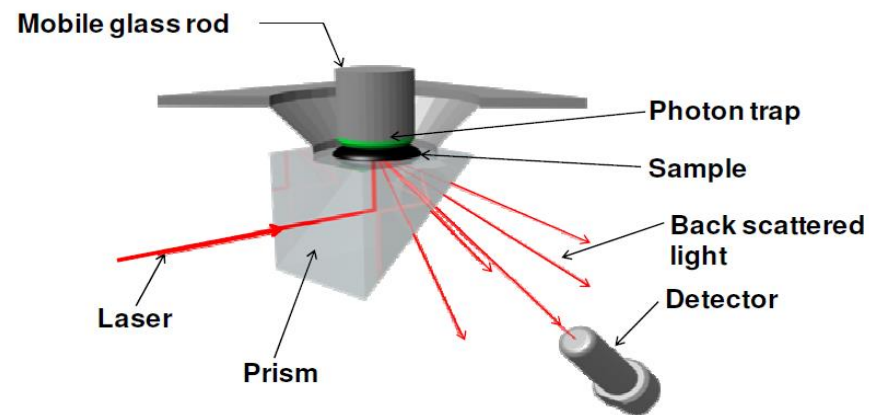
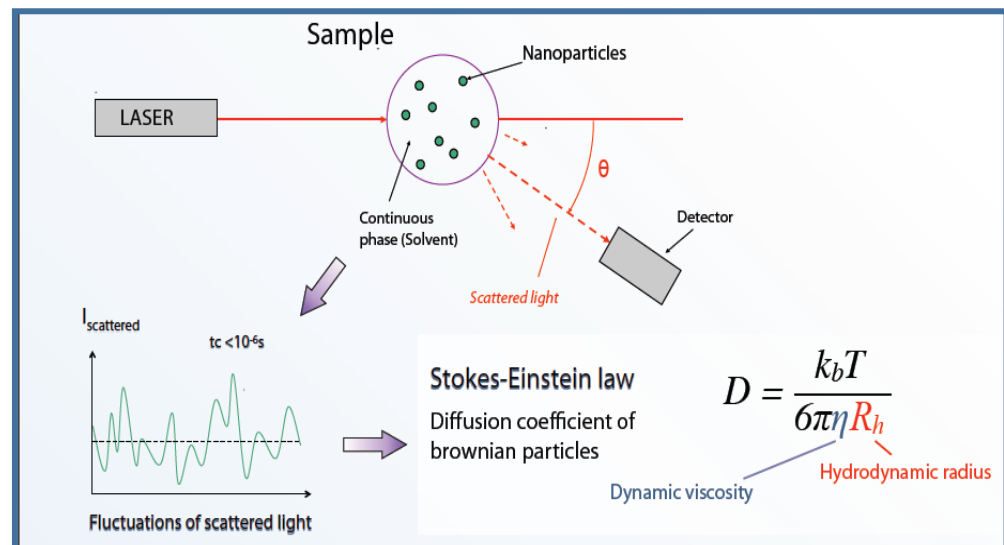
Water-in-Crude Oil Emulsions

Al Ghamdi *et al.* (2009), *SPE Journal* 14(4), 595.

CHARACTERIZATION OF EMULSIONS – DROPLETS SIZE

Dynamic Light Scattering (DLS): powerful technique to characterize diluted and transparent dispersions of particles from the **nanometer up to a few microns** through the analysis of scattered light fluctuations caused by the Brownian motion of particles.

To overcome the DLS limitations, IFPEN has developed an original optical design of the sample cell. The bottom of the measuring cell is formed by the upper surface of a glass prism guiding the laser beam while a mobile glass rod allows an accurate control of the sample thickness (down to a few tens of microns thickness).



Vasco DLS instrument (Cordouan Technologies)



Yudin et al. (1998), *Journal of Petroleum Science and Engineering* **20**, 297.

CHARACTERIZATION OF EMULSIONS – ELECTRICAL STABILITY



Dalmazzone *et al.* (2010),
SPE Journal 15, 726.

Electrical Stability Tester EST

Method and electrical apparatus used for oil-based drilling mud stability control (API 13B-2)

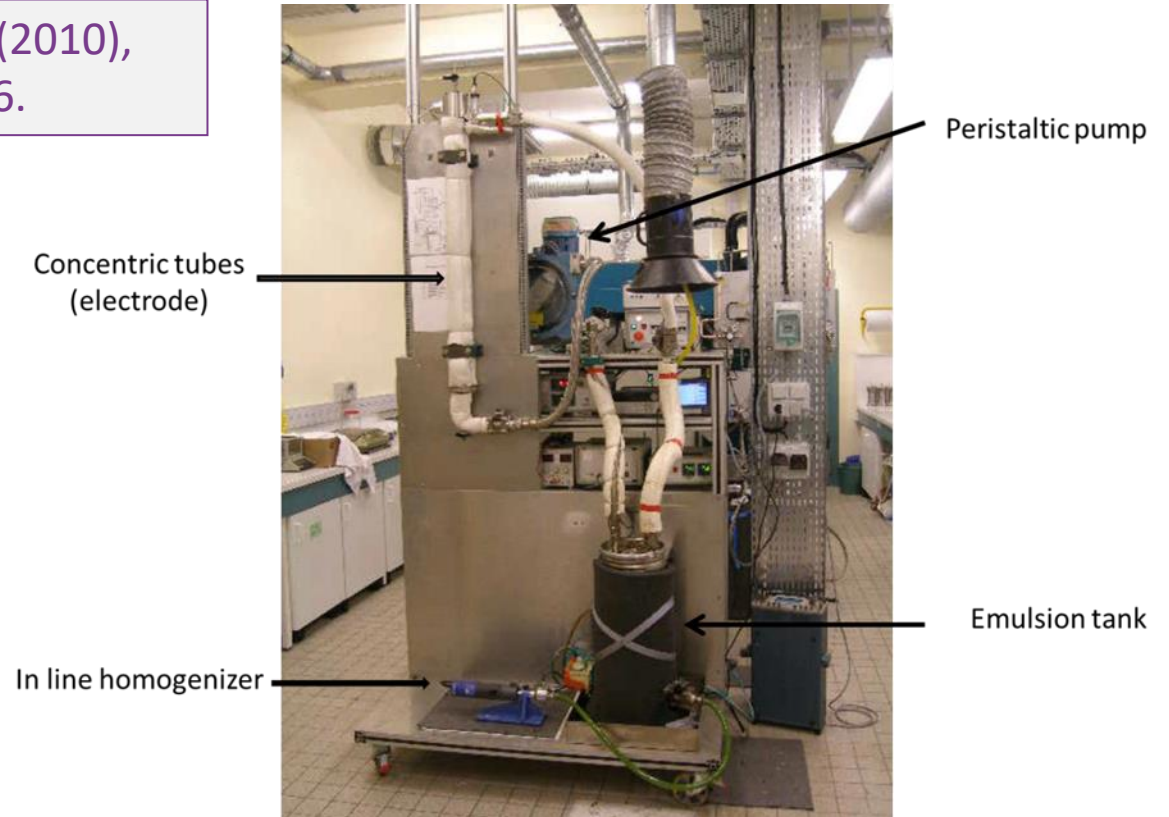
Simple electrode, AC Voltage @ 340 Hz

Progressive increase of voltage @ 150 V/s up to 12.9 kV/cm

Critical Electrical Voltage measured at 61 μ A (short-circuiting)

Method: Tests on reconstituted emulsions

After EST test, monitoring of the volume of separated free water after 24 hours



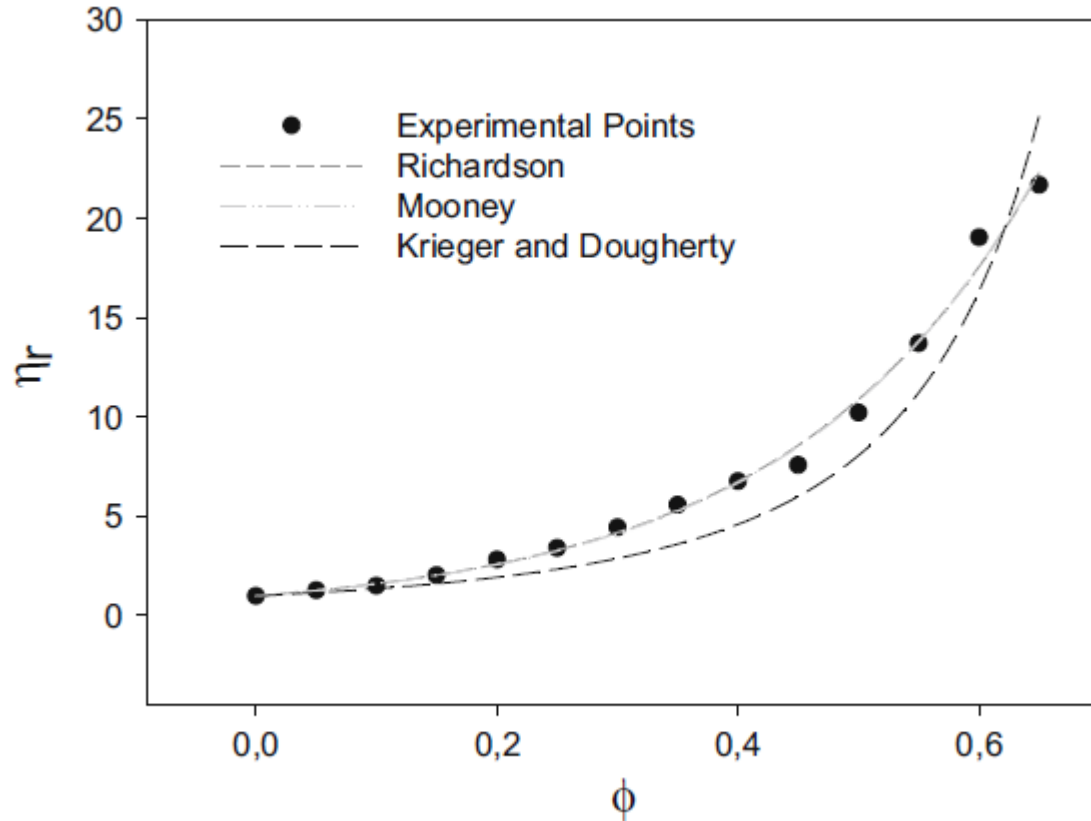
IFPEN Electrocoalescence set-up

Fluid injection into the annular between 2 concentric tubes as electrodes

AC field: high voltage/high frequency 50Hz – 2 kHz

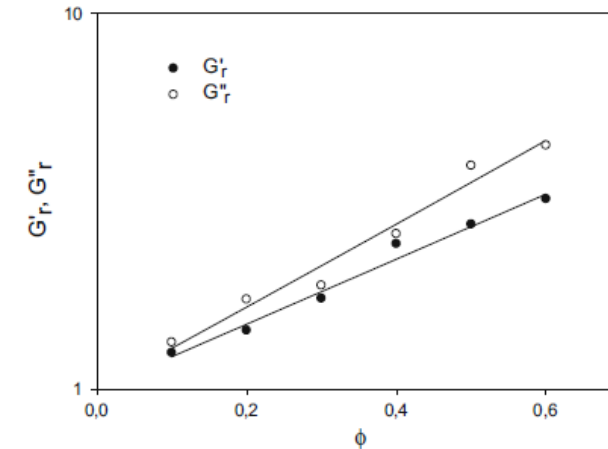
Flow rate 0.1 – 1.2 m³/h - T from ambient to 60°C

CHARACTERIZATION OF EMULSIONS – RHEOLOGY

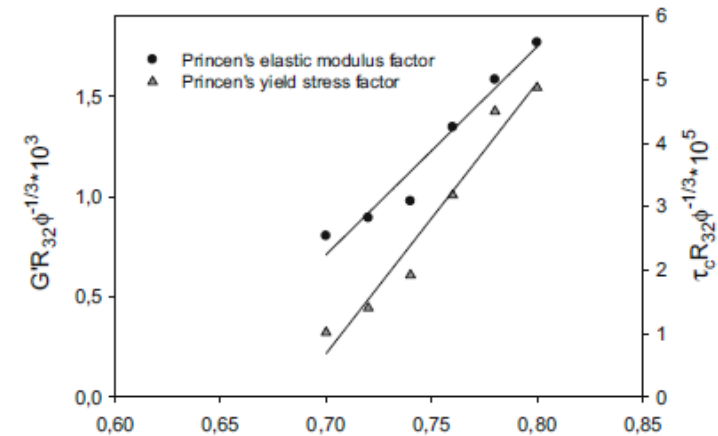


Comparative plot of different viscosity models (relative viscosity versus dispersed volume fraction) – 30°C

Quintero, Noik, Dalmazzone and Grossiord (2008), *Rheol Acta* 47, 417.



G' and G'' vs volume fraction, 20°C, 1 Hz



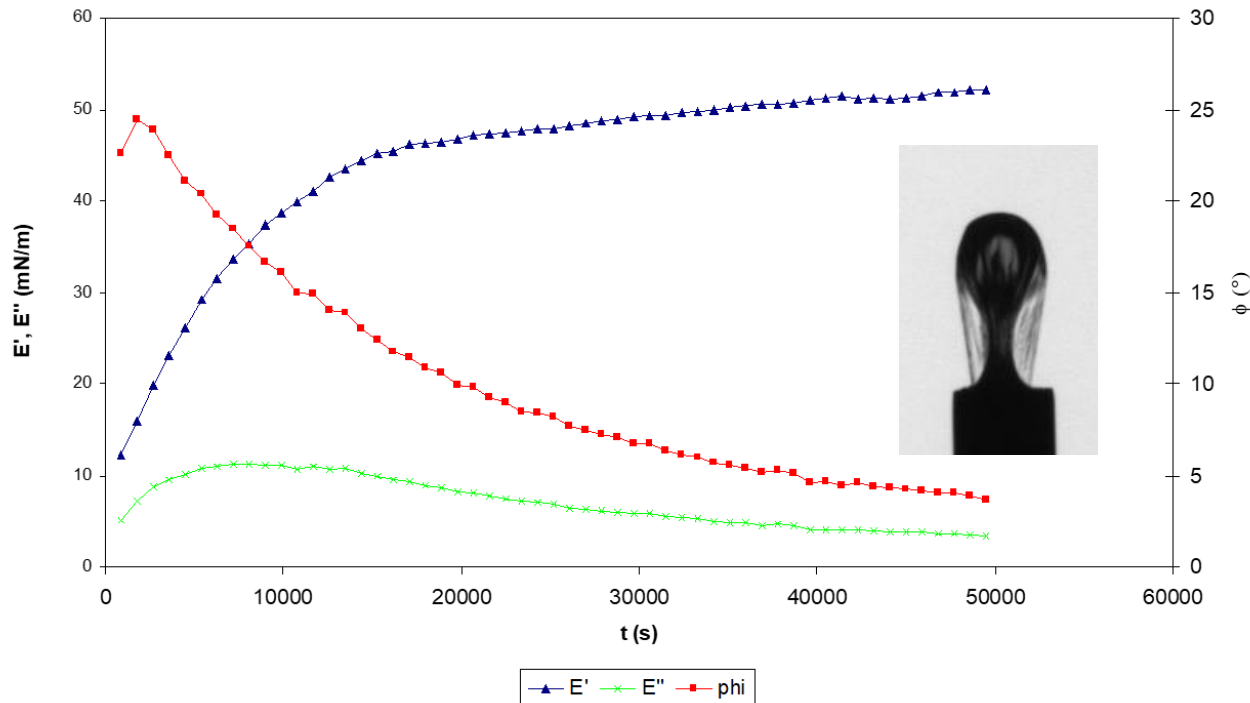
Princen parameters for G' vs volume fraction, 20°C

CHARACTERIZATION OF EMULSIONS – INTERFACIAL RHEOLOGY

Tracker (TECLIS) for interfacial rheology



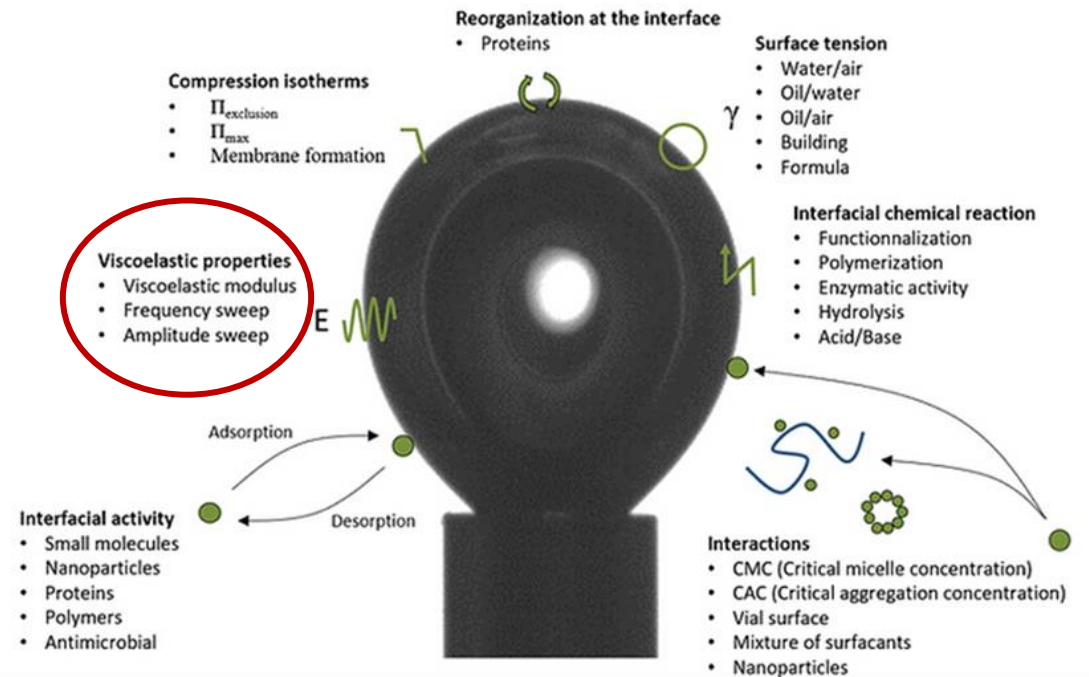
Evolution of E' , E'' and ϕ with time (crude C, 0.1 Hz, 30°C)



Quintero, Noik, Dalmazzone and Grossiord (2009), *OGST-Rev. IFP* 64, 607.

Pendant or Rising drop Measurements overview

Source: Teclis



CONCLUSION

- The step of characterization is essential at the lab and semi-industrial scale
 - To develop flexible and efficient technologies
 - To develop specific additives (emulsion breakers, flocculant, clarifiers...)
- Combination of several methods to characterize opaque and concentrated emulsions
 - Stability
 - Droplets size
 - Rheological behavior

Acknowledgments: C. Noïk, A. Mouret, L. Podesta-Foley, P. Thoral, C. Burnichon, D. Frot, I. Hénaut

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 www.ifpenergiesnouvelles.fr

 @IFPENinnovation

