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AgroParisTech
Talents d'une planète soutenable

SayFood

Monitoring the degradation status of biodegradable polymers by assessing thermal properties

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Université de Montpellier – INRA – Montpellier - France*

A tremendous demand for plastics



Plastic wastes accumulation => environmental damage

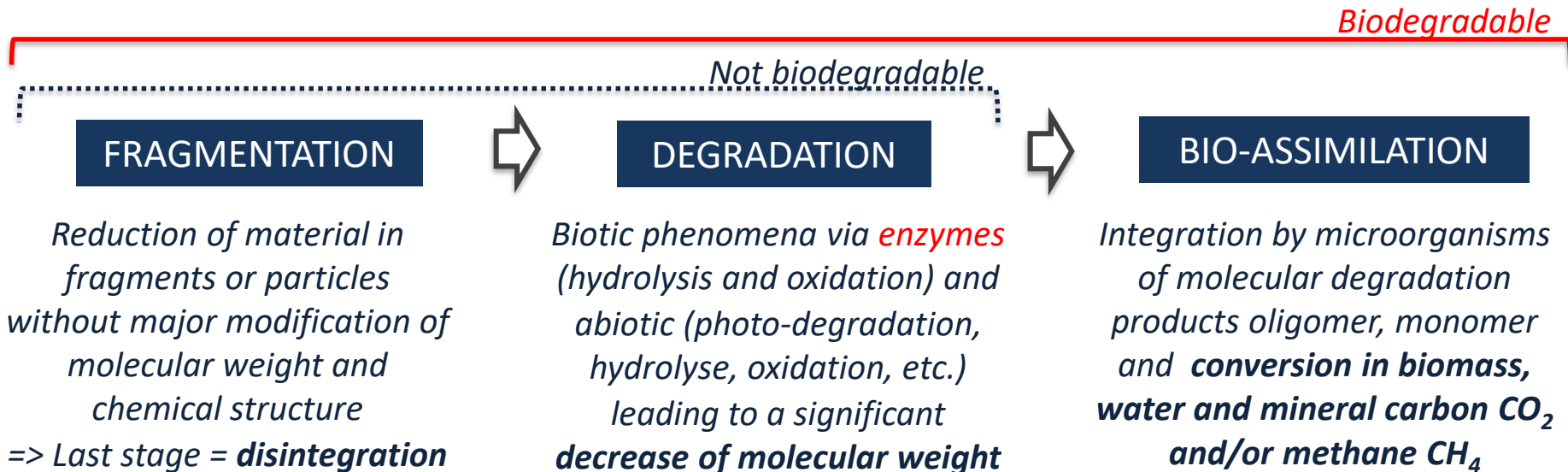
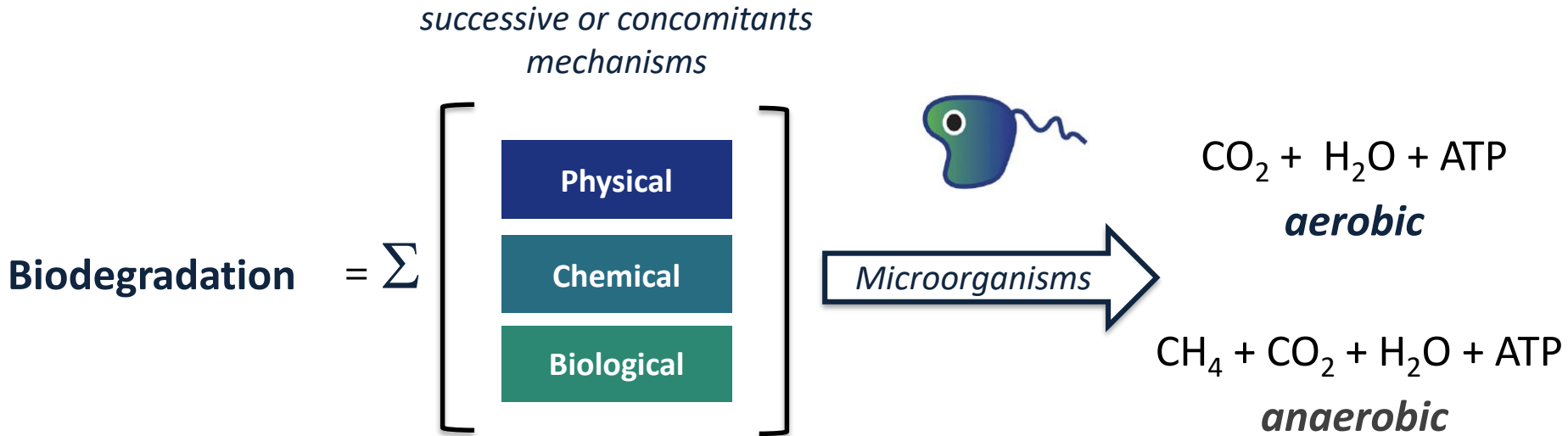


Terrestrial



Marine

Mechanisms involved in biodegradation



MACROSCOPIC

MICROSCOPIC

MACROMOLECULAR

MOLECULAR

Understanding of the main **physical-chemical properties**
that affect their biodegradation rate

to

Identify **key limiting factors** driving biodegradation



by combining different methodological approaches

PHBV

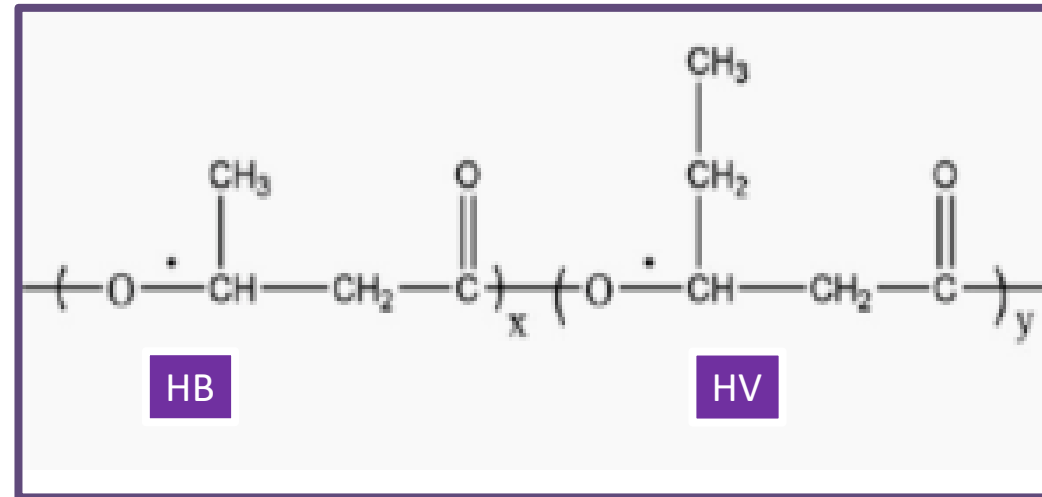
Polyhydroxybutyrate-co-valerate

PHI 002 (NaturePlast)

1-3% mol HV

Xc = 54 % (DSC)

Mn ≈ 94 000 g/mol



PBSA

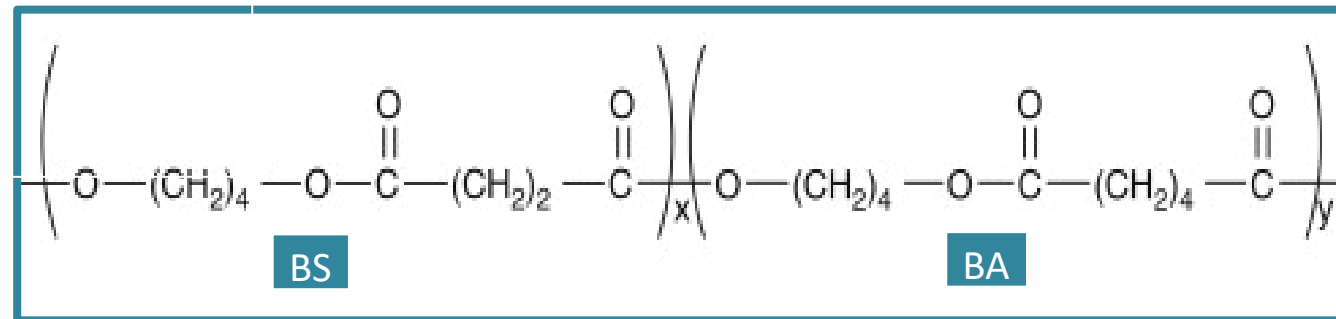
Polybutylene succinate-co-adipate

PBE 001 (NaturePlast)

21% mol BA

Xc = 47% (DSC)

Mn ≈ 44 000 g/mol



PLA

Poly(lactic acid)

Natureworks

Amorphous

d = Ø

Mn ≈ 75 311 g/mol

Total Corbion

Semi-crystalline

d = 31.13 % (DSC)

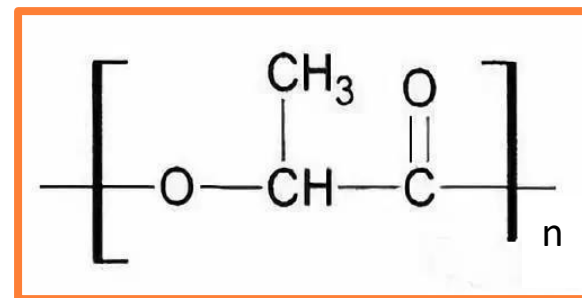
Mn ≈ 82 960 g/mol

Lyspackaging

Semi-crystalline

d = 25.52 % (DSC)

Mn ≈ 103 760 g/mol



PHBV

PBSA

**Biodegradation process
in composting conditions**



Pieces of film



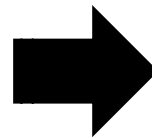
100 days



50% humidity



58°C



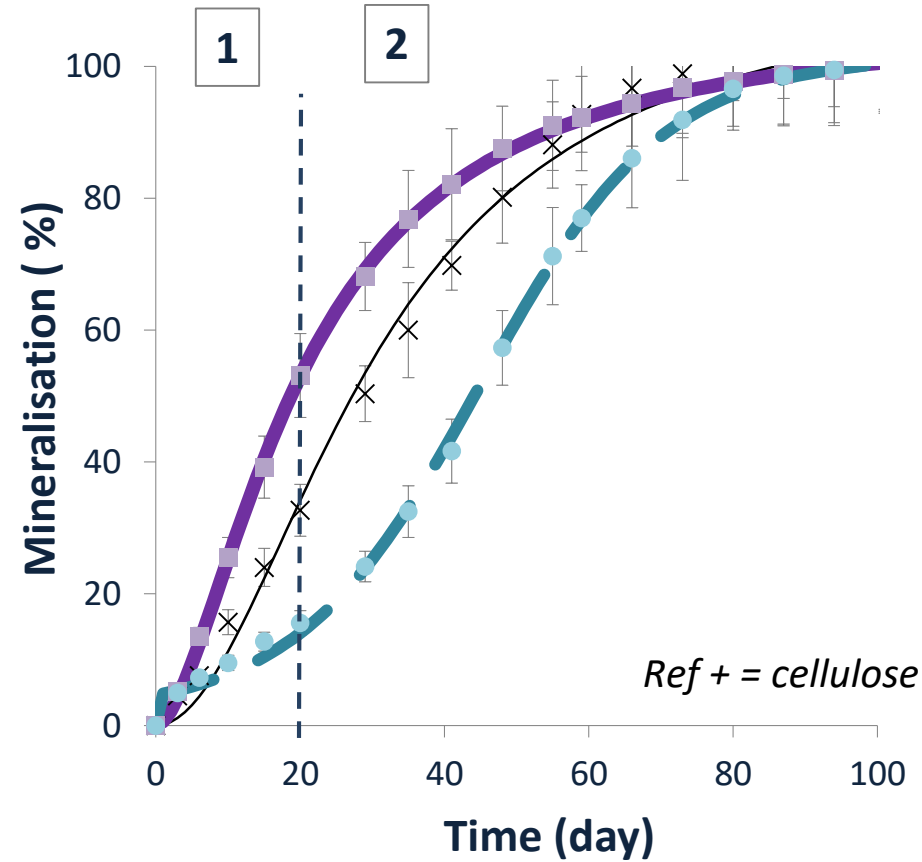
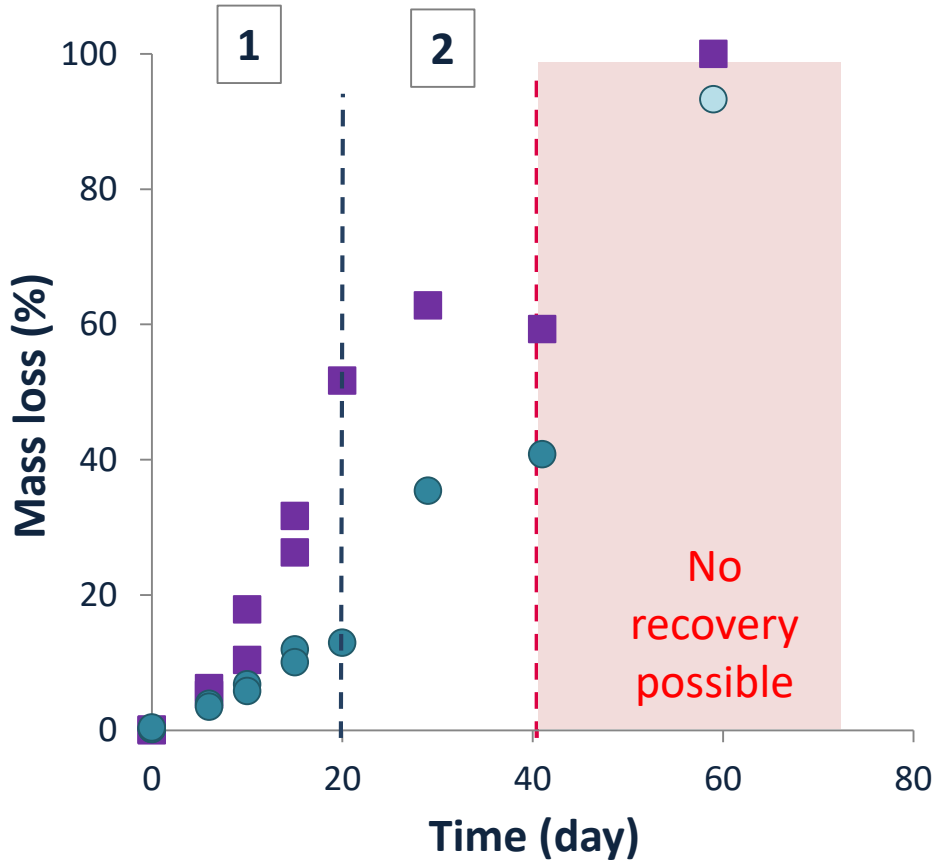
sampling during
the
biodegradation
process

Physical-Chemical Analysis

- Mass loss (weighting)
- Mineralisation (respirometry)
- Surface erosion (SEM)
- Surface hydrolysis (FTIR)
- Molar Mass (SEC)
- Tg, Xc (DSC)

Biodegradation in industrial composting conditions (58°C)

PHBV
PBSA



Phase 1 : PHBV > PBSA
Phase 2 : PHBV < PBSA

PHBV

vs

PBSA

Degradation time (day)

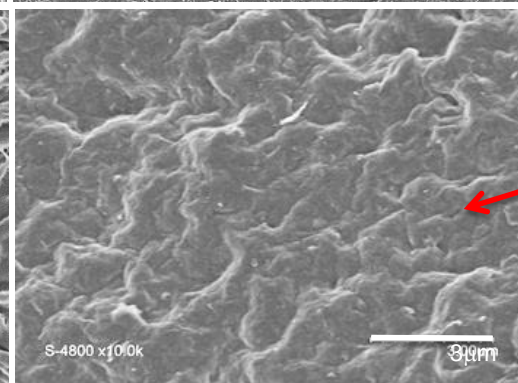
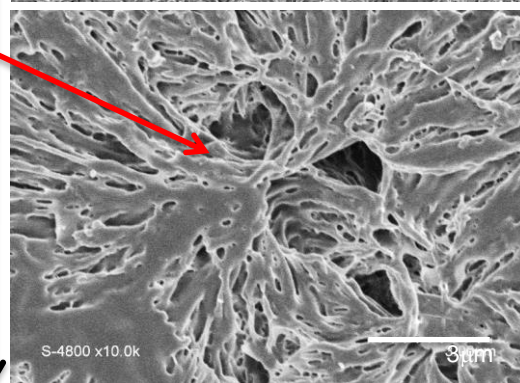
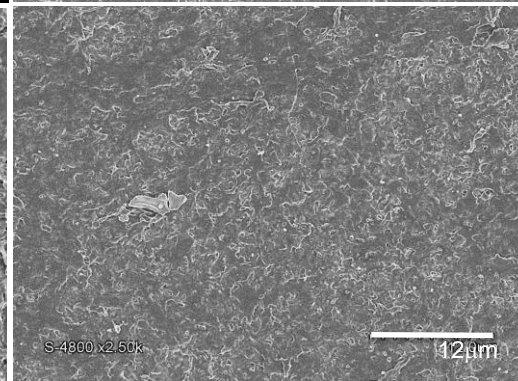
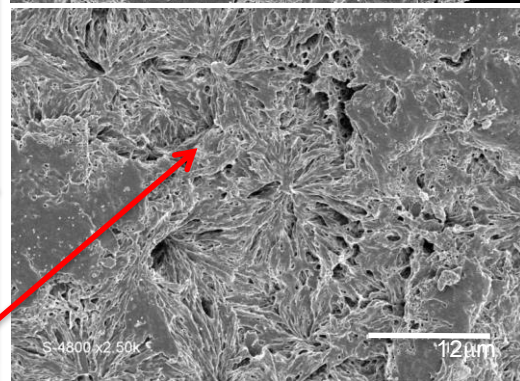
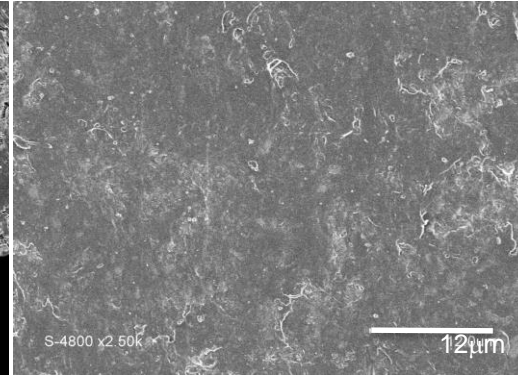
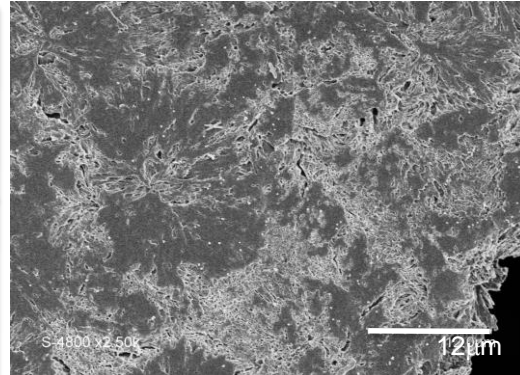
6

Significant erosion/etching

10

spherulites

20

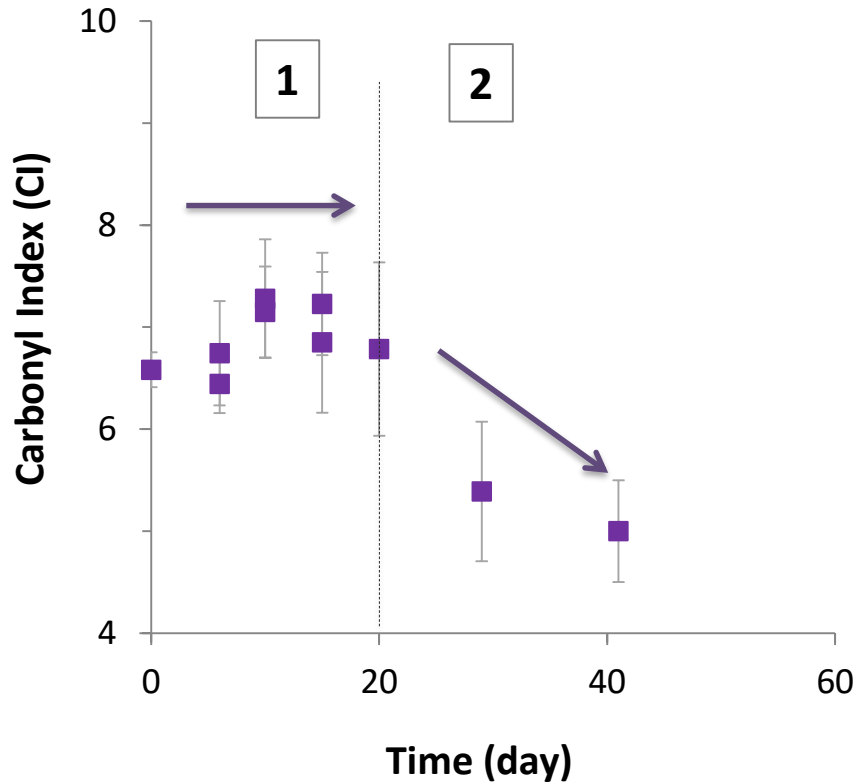


flat layers of crystal

Erosion delayed

PHBV

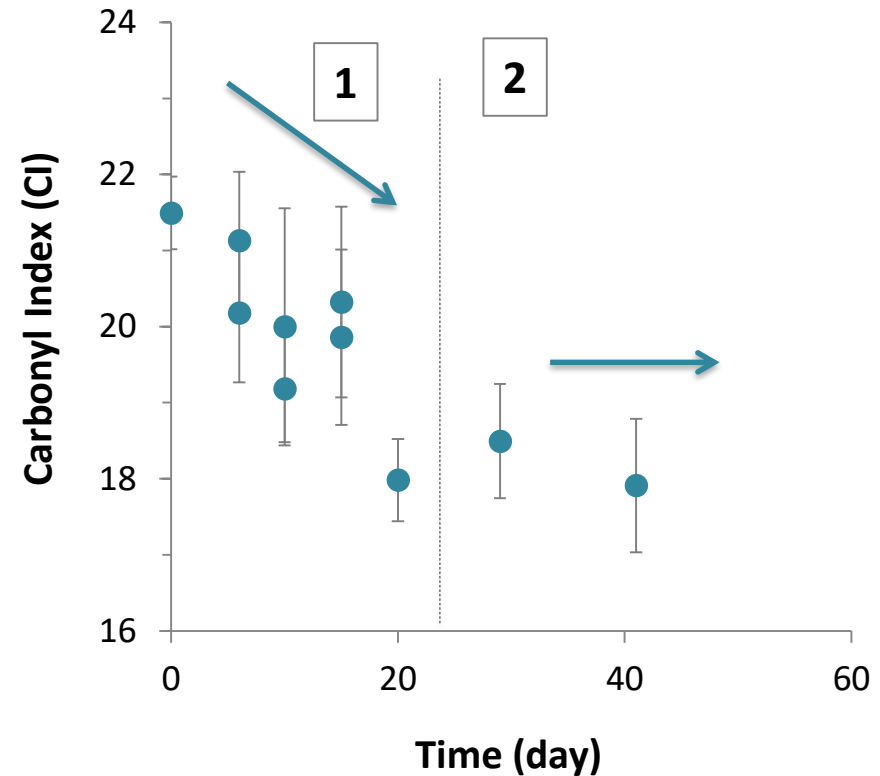
$$CI = \frac{\text{Abs } 1770\text{-}1700 \text{ cm}^{-1} (\text{C=O})}{\text{Abs } 1379 \text{ cm}^{-1} (\text{REF } \text{-CH}_3)}$$



PBSA

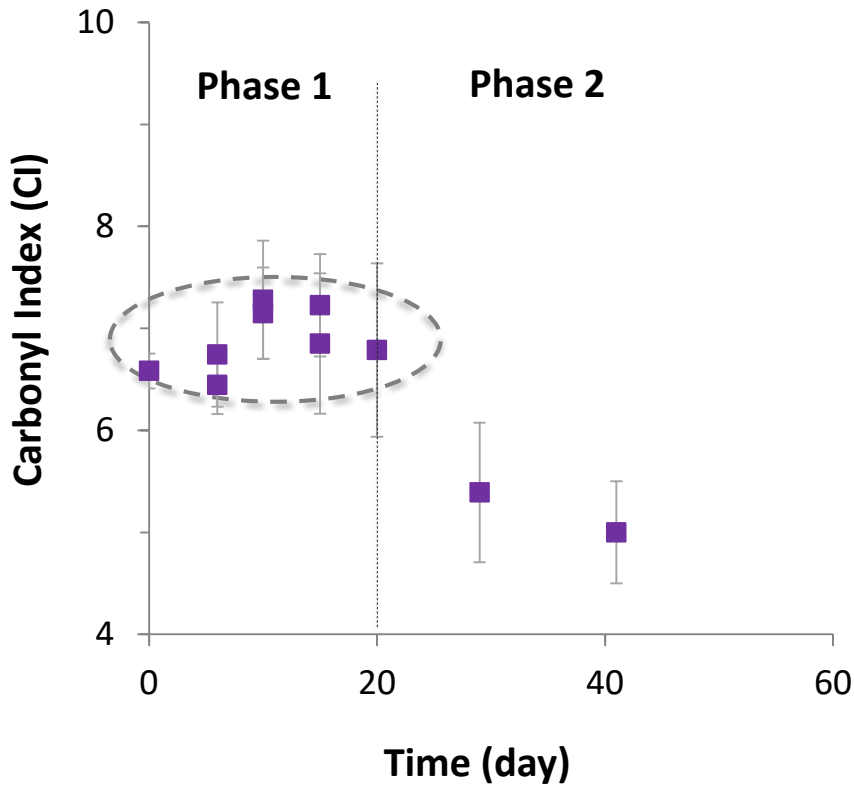
FTIR

$$CI = \frac{\text{Abs } 1770\text{-}1690 \text{ cm}^{-1} (\text{C=O})}{\text{Abs } 1473 \text{ cm}^{-1} (\text{REF } \text{-CH})}$$



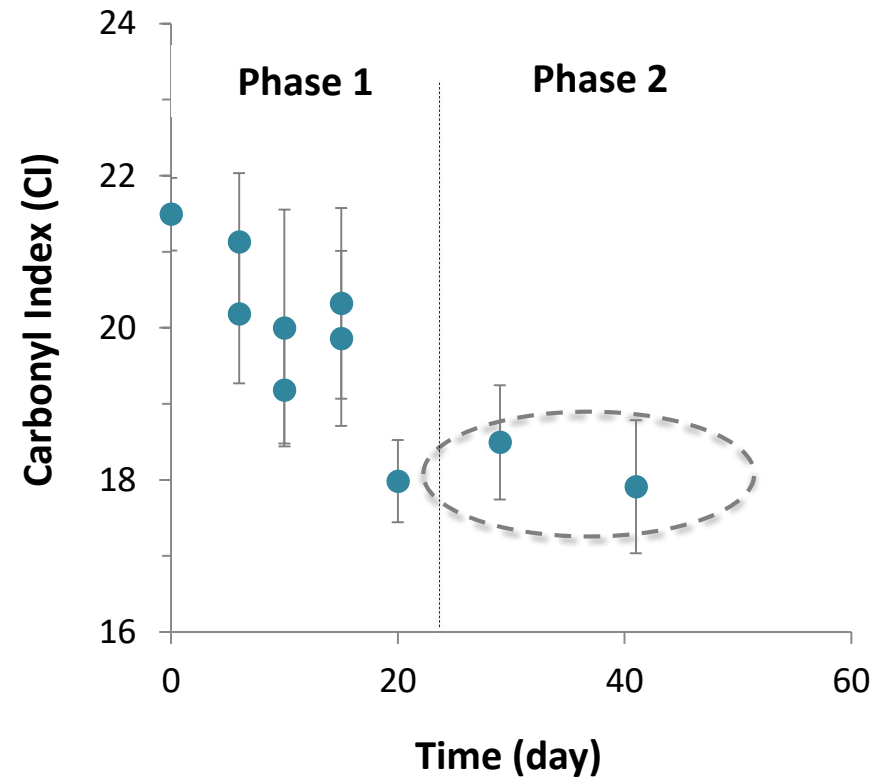
✓ Carbonyl Index => indicator of the extent of the ester linkage hydrolysis

PHBV



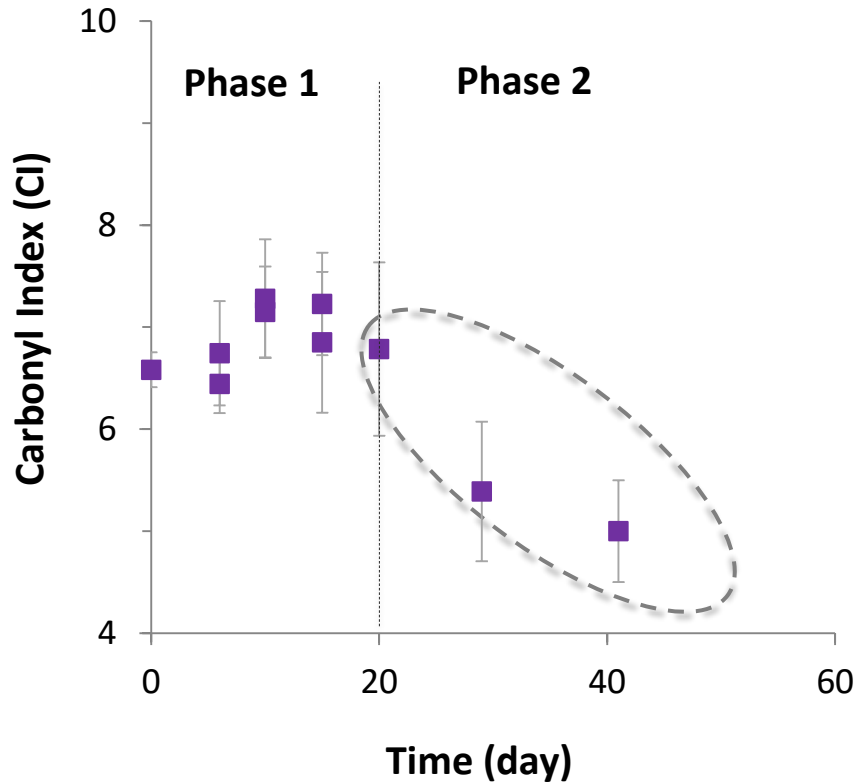
PBSA

FTIR

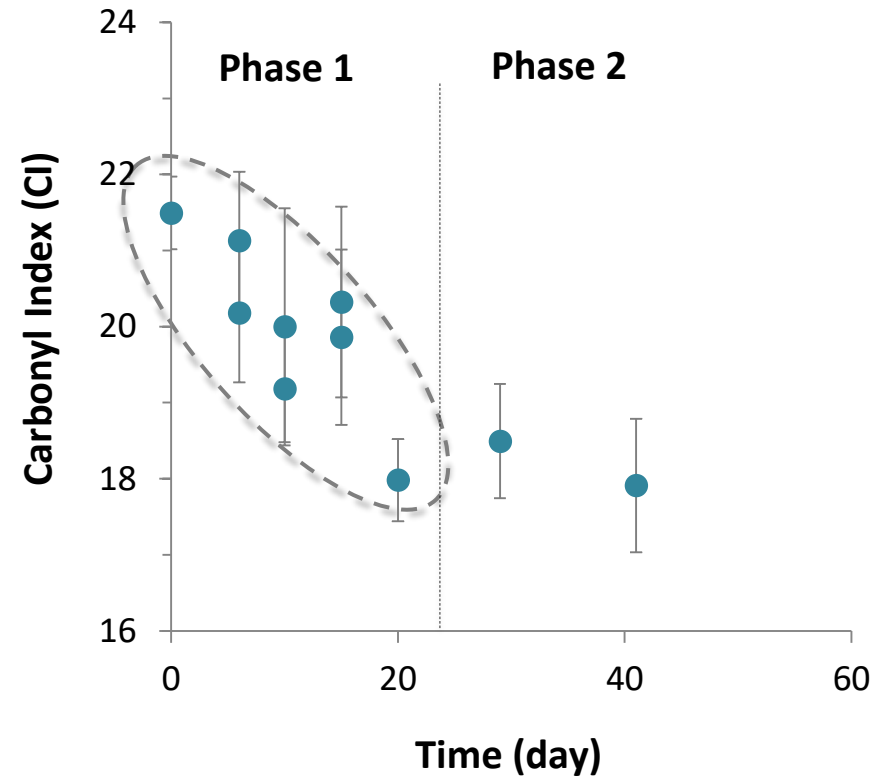


- ✓ Hydrolysis concomitant with erosion => **no change in the carbonyl index**
- ✓ Related with a **high rate** of mass loss and mineralisation

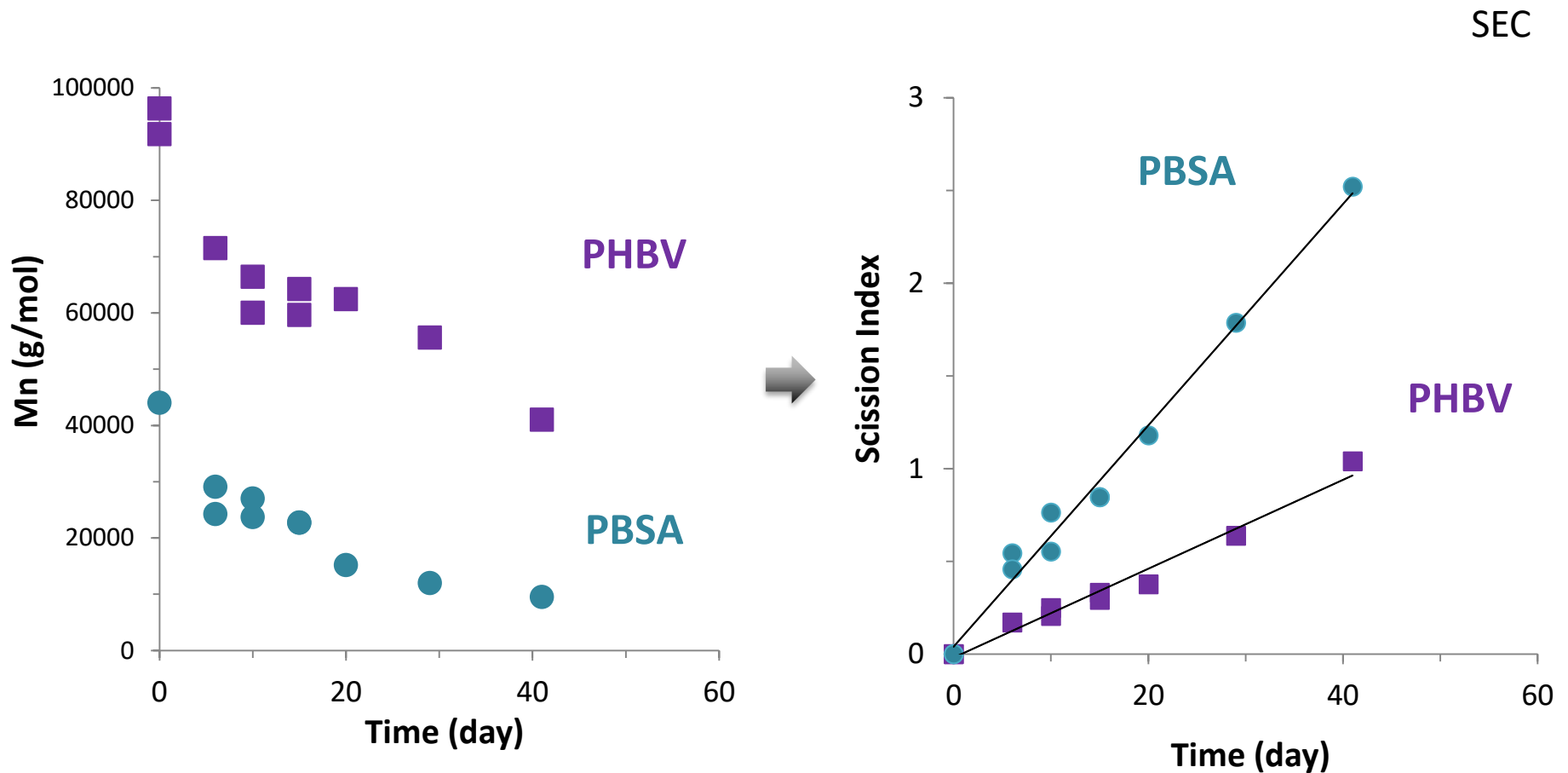
PHBV



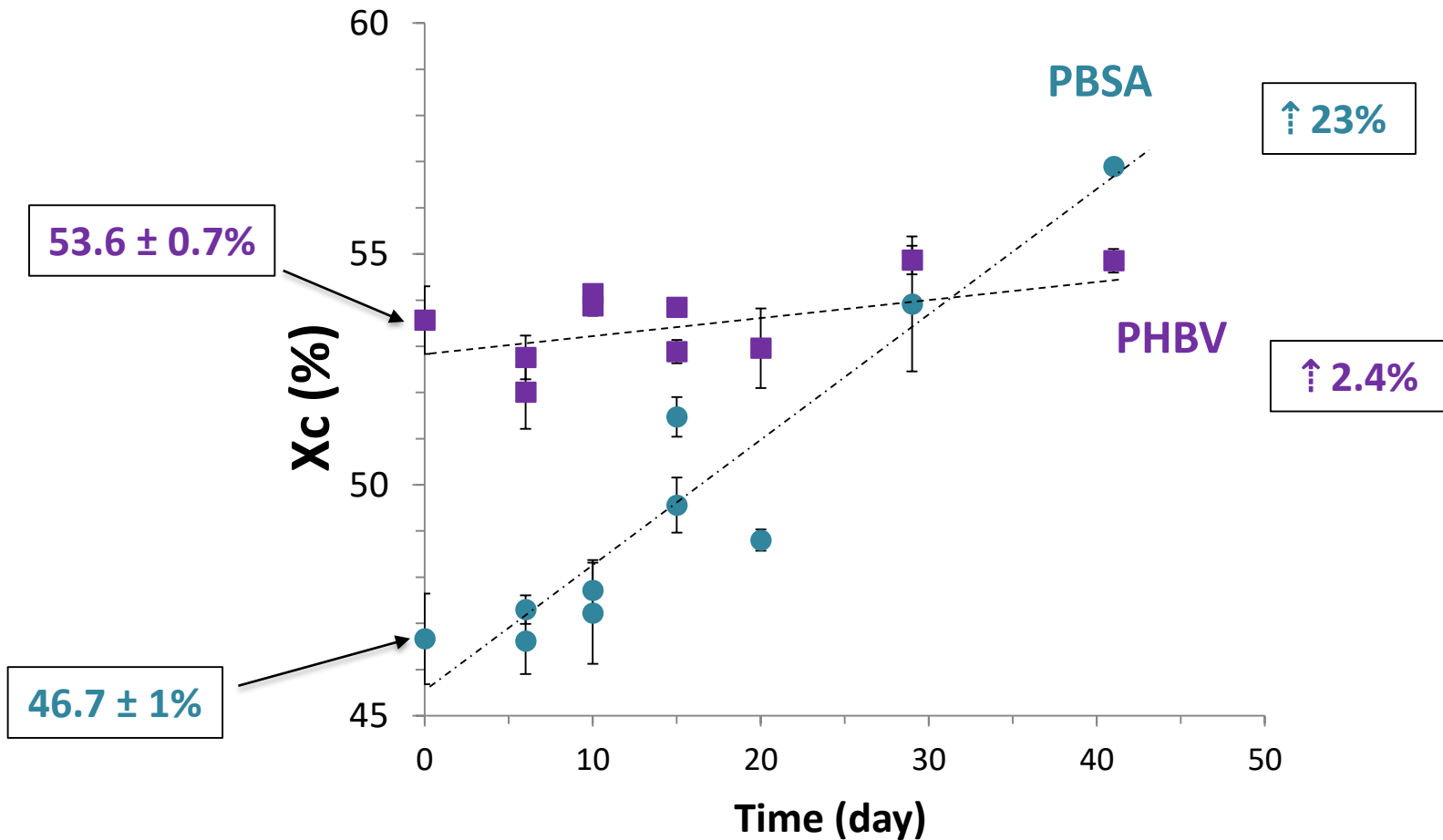
PBSA



- ✓ Hydrolysis **without** erosion/etching => **decrease of the carbonyl index**
- ✓ Related with a **low rate** of mass loss and mineralisation

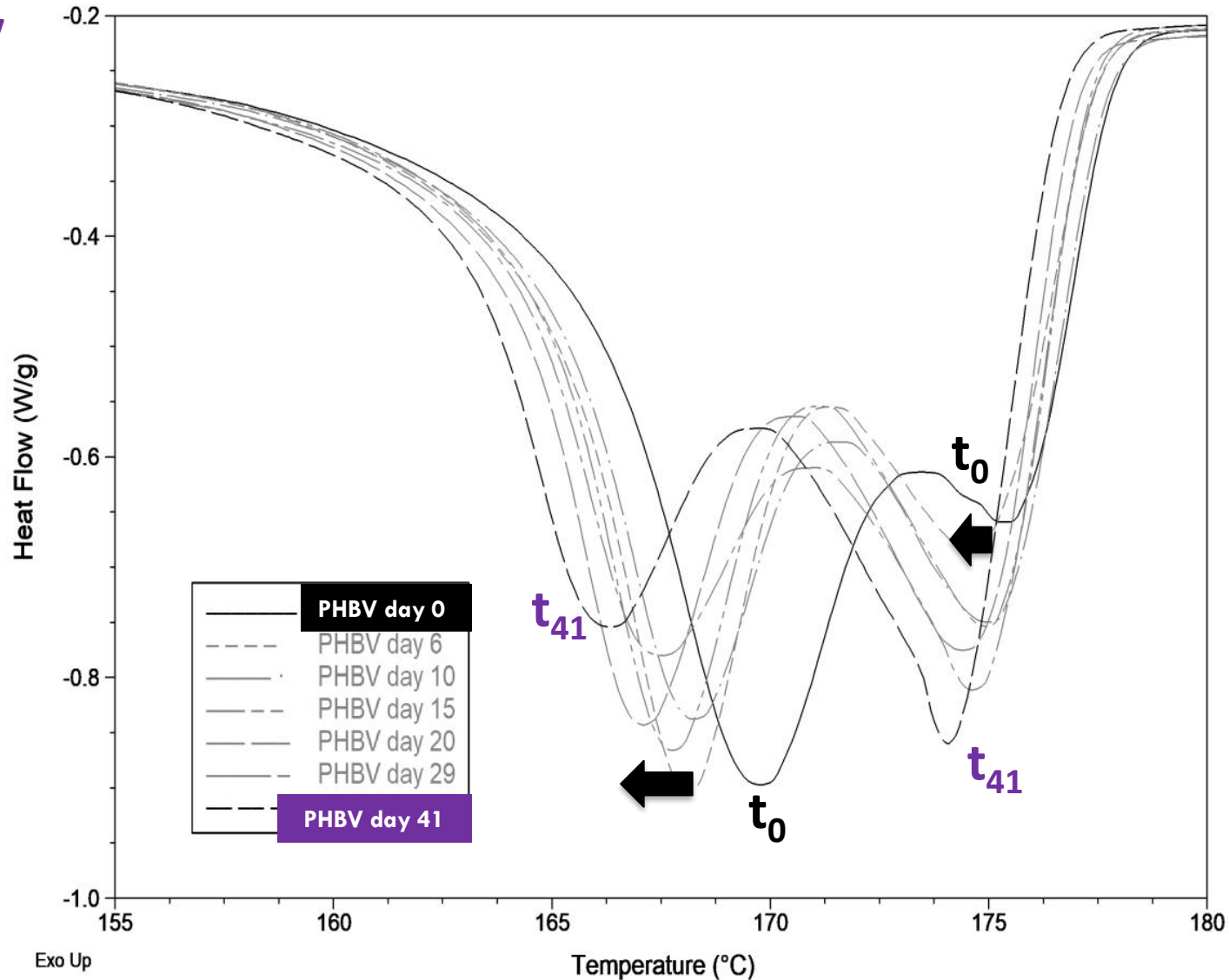


- ✓ Hydrolytic chain scission **PBSA > PHBV**
- ✓ Related to **water permeability and diffusion** => **abiotic** mechanism



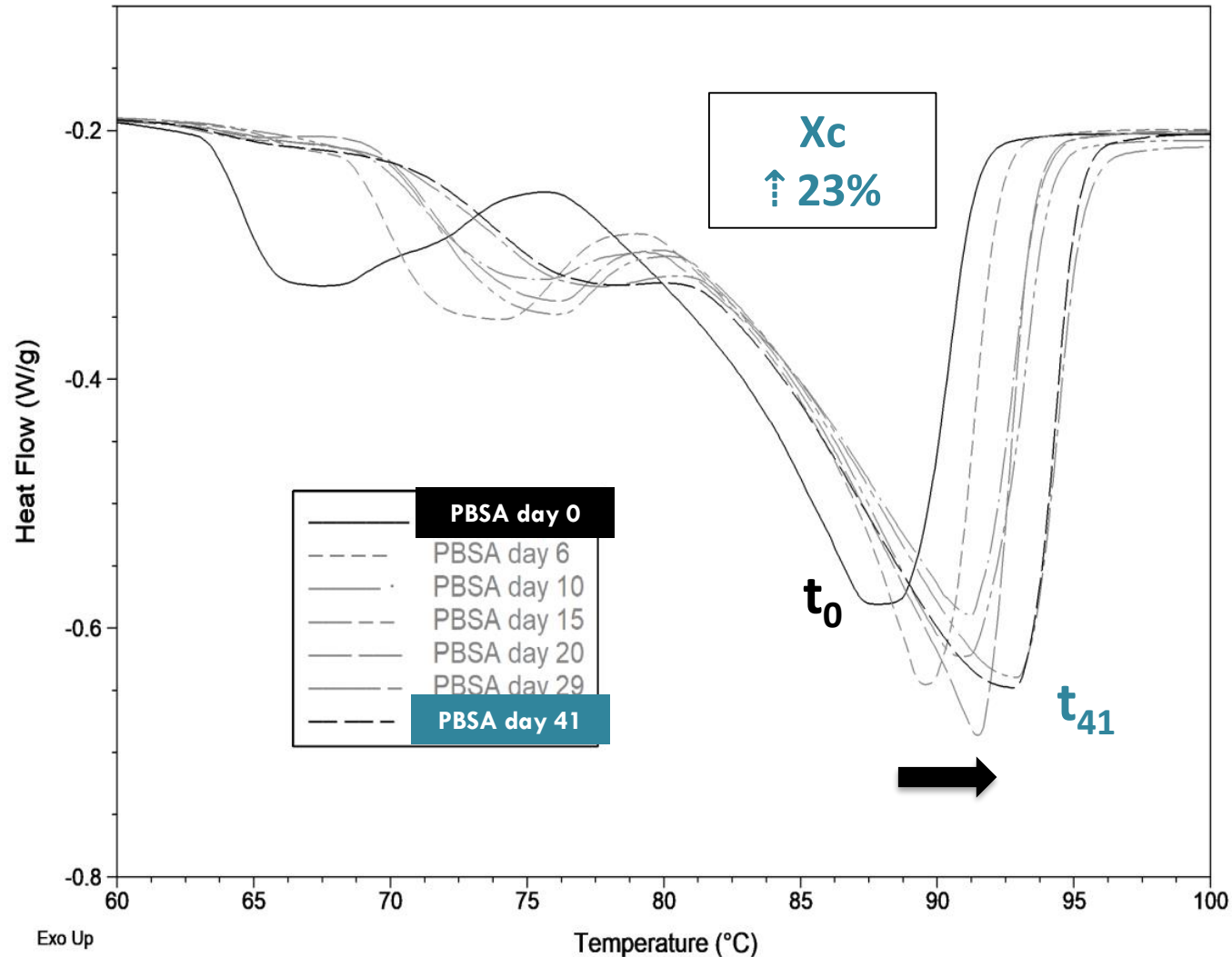
- ✓ **Hyp 1** : ≠ accessibility of the crystalline phase for micro-organisms enzymes
PHBV >> PBSA due to ≠ morphologies and specific surface
- ✓ **Hyp 2** : recrystallisation phenomenon due to water hydrolysis of **PBSA** chains

PHBV



- ✓ new crystal populations with different morphologies and dimensions
- ✓ reduction in molecular weight of PHBV polymer chains.

PBSA



- ✓ Increase in T_m reflecting a lower flexibility of polymer chains
- ✓ Increase of X_c reflecting an increase in the level of structuration of the crystals
- ✓ Behaviour in link with the **ability of PBSA to fragment** during degradation in compost

PLA

Pellets

amorphous

semi-crystalline ($X_c = 31\%$)

Packaging material

semi-crystalline ($X_c = 25\%$)

LYS VEGANBOTTLE
PACKAGING
DESIGN · ENGINEERING · PRODUCTION



Thermo-chemical treatments



Pellets
Pieces of bottle



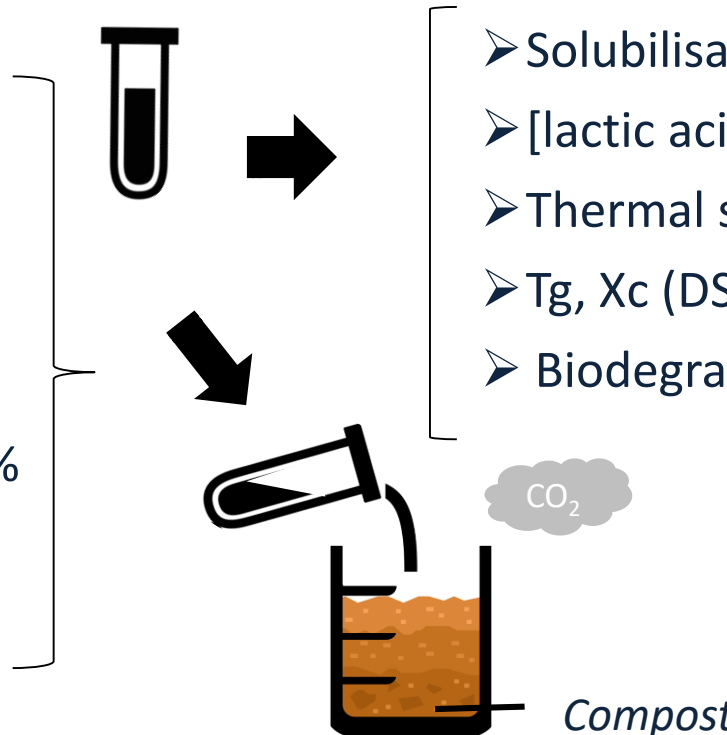
24 à 48h



Water / KOH 2%, 20%



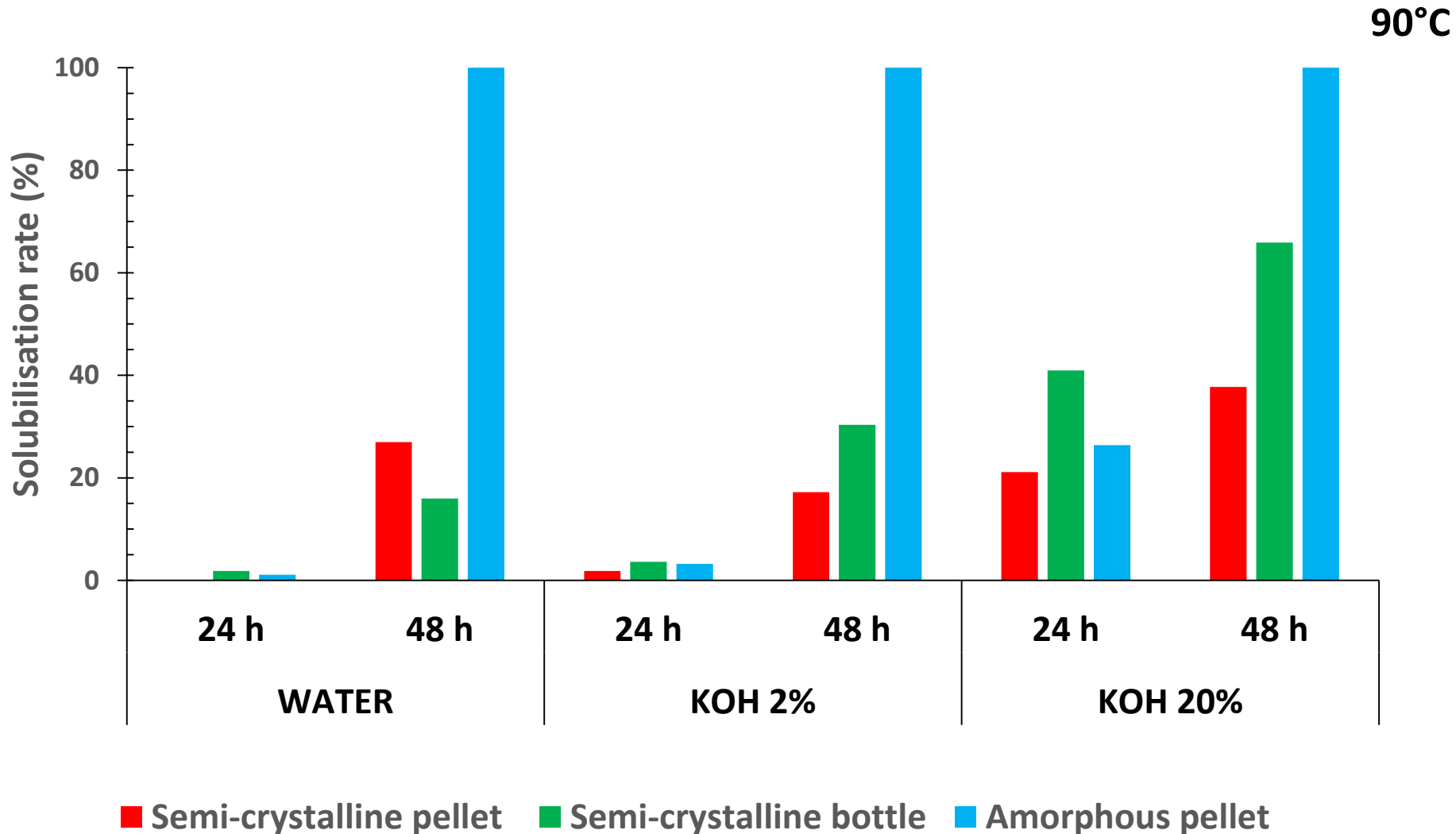
70-90-100°C



Physical-Chemical Analysis

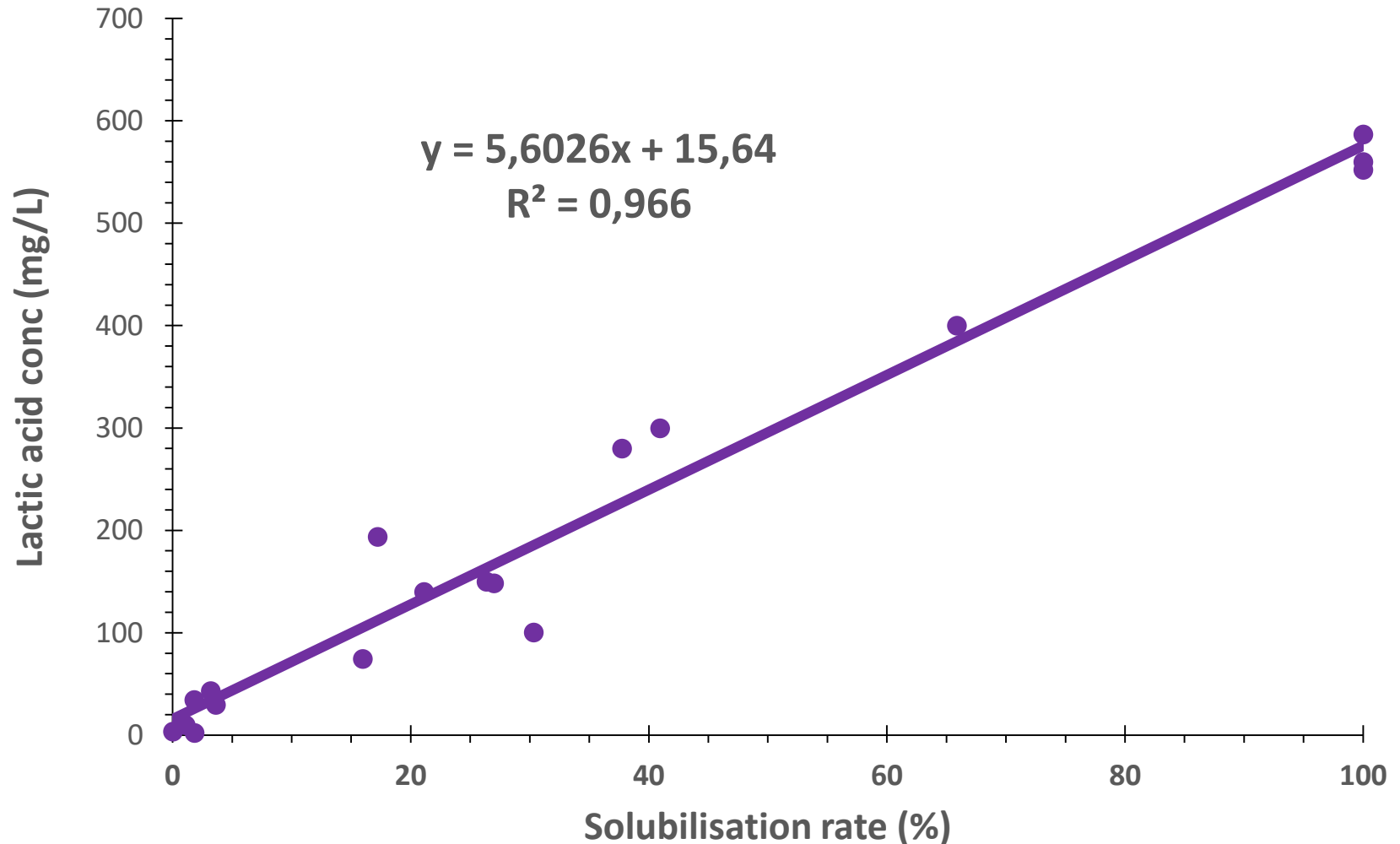
- Solubilisation rate (weighting)
- [lactic acid] (HPLC)
- Thermal stability (TGA)
- T_g , X_c (DSC)
- Biodegradation (respirometry)

Impact of thermo-chemical treatment on PLA solubilisation



- ✓ Solubilisation rate of PLA is related to polymer crystallinity
- ✓ Solubilisation rate of PLA increases with the duration of the treatment

Relation between solubilisation and hydrolysis



✓ Good correlation between solubilisation and hydrolysis whatever the degree of crystallinity and type of treatment

TGA

N₂

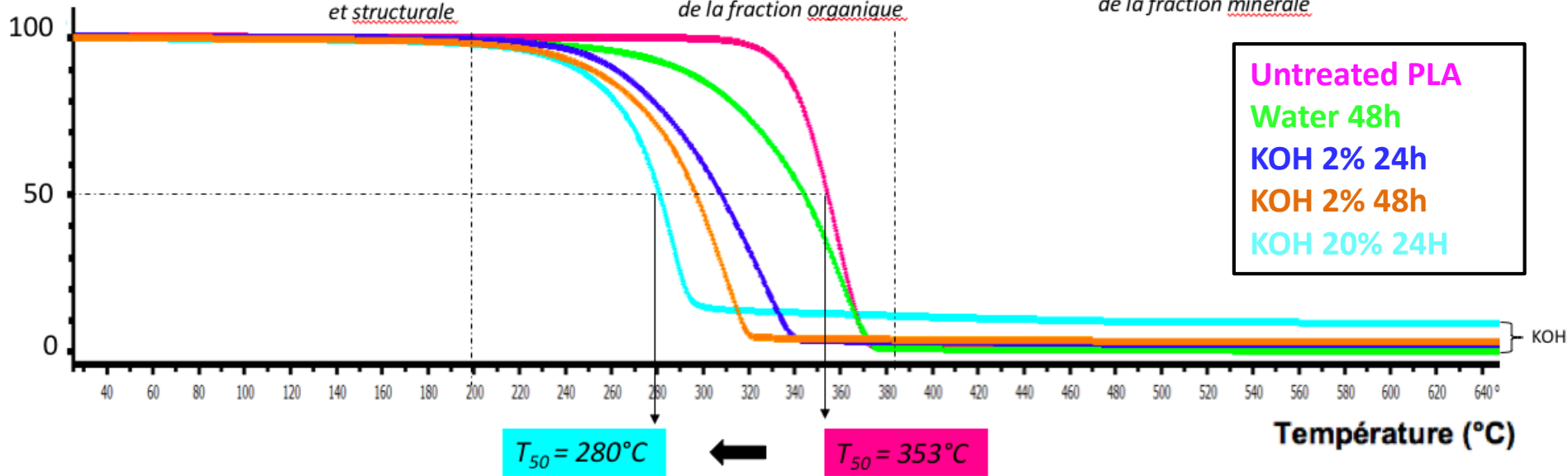
Semi-crystalline bottle

Mass Loss (%)

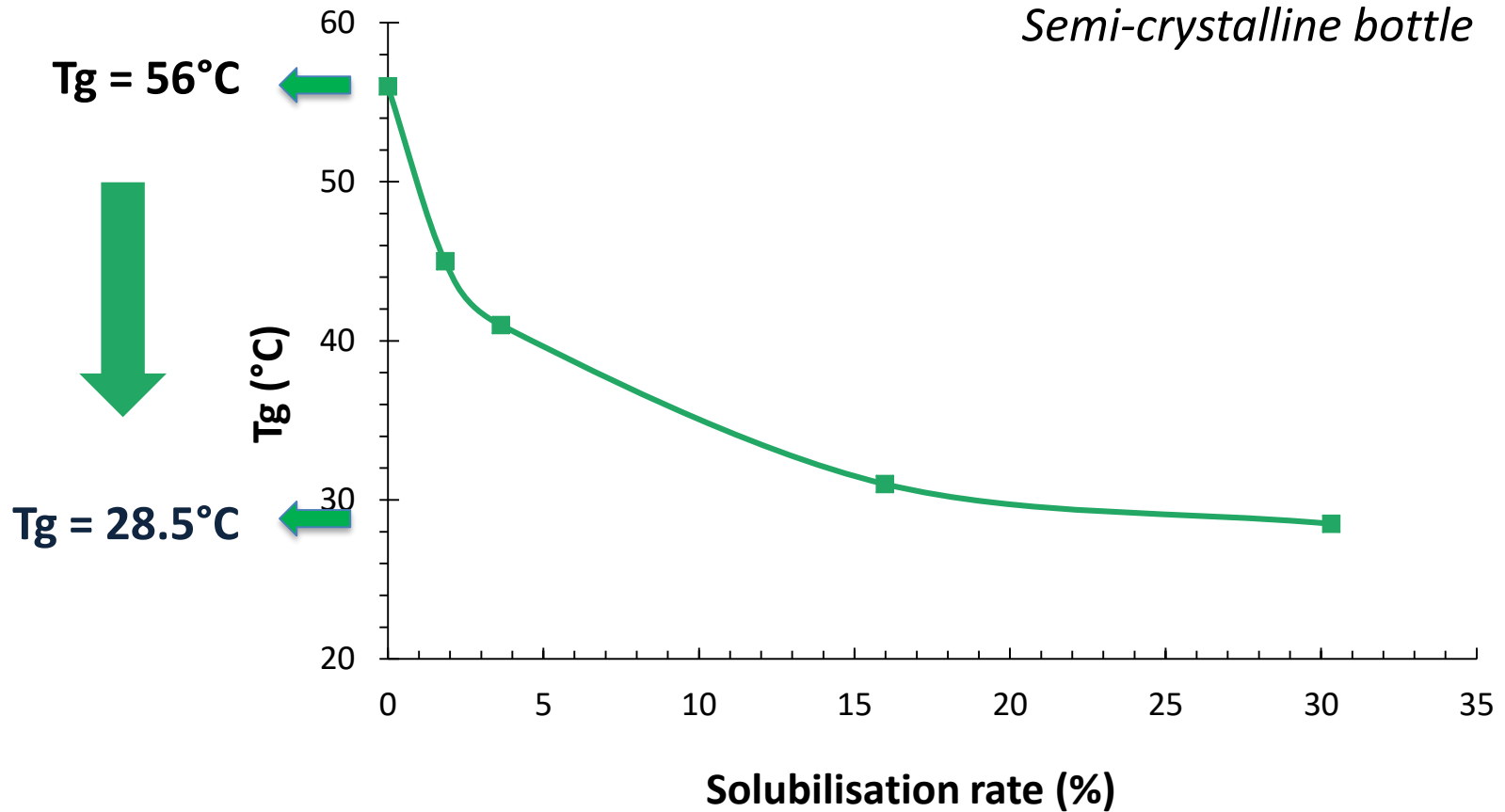
I Perte eau libre et structurale

II Oxydation thermique de la fraction organique

III Oxydation thermique de la fraction minérale



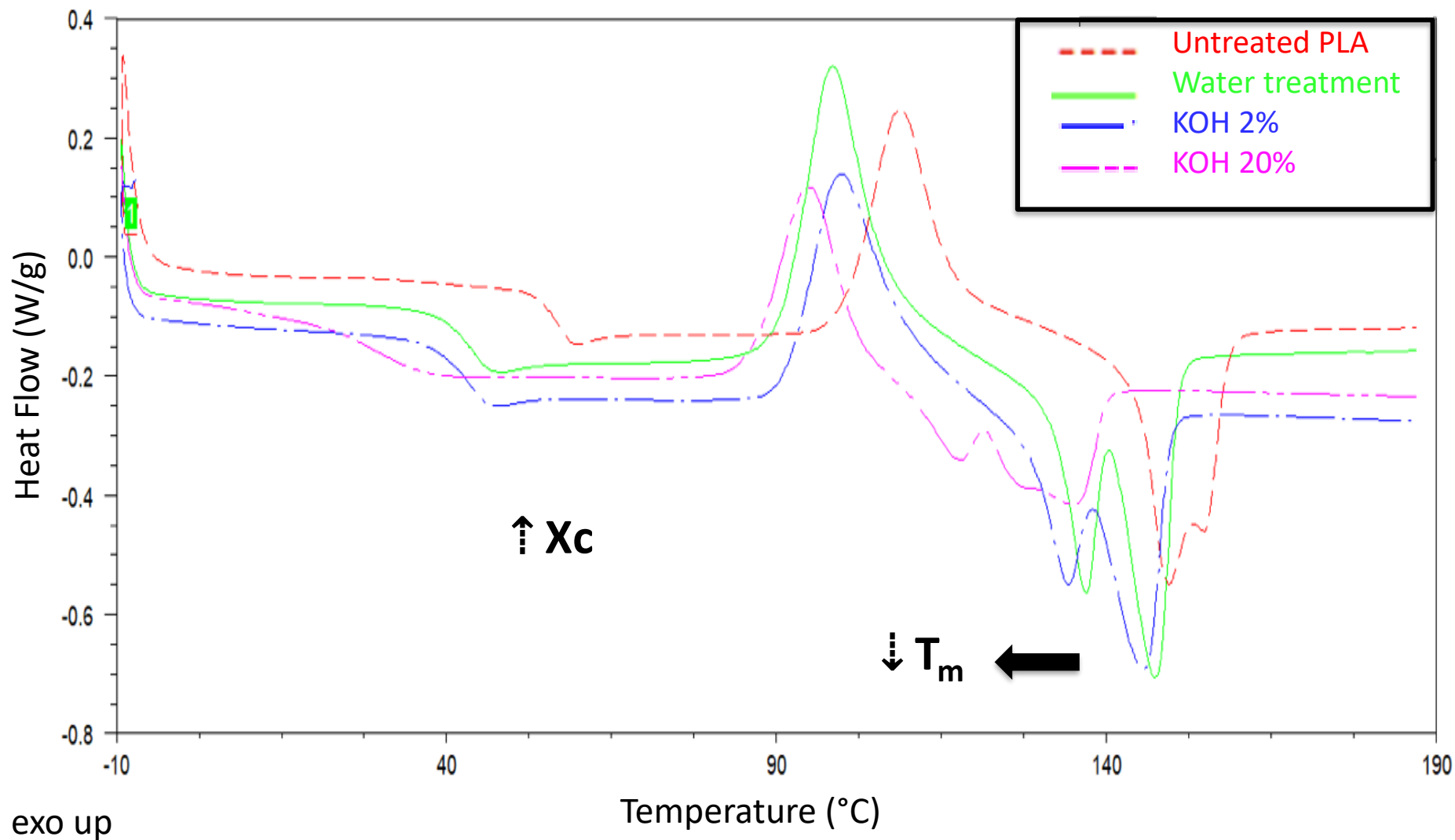
✓ Solubilisation rate \Leftrightarrow thermal stability



↗ Solubilisation rate \Leftrightarrow ↗ Hydrolysis rate \Leftrightarrow ↘ Tg
↗ Chain mobility with decrease in Mw

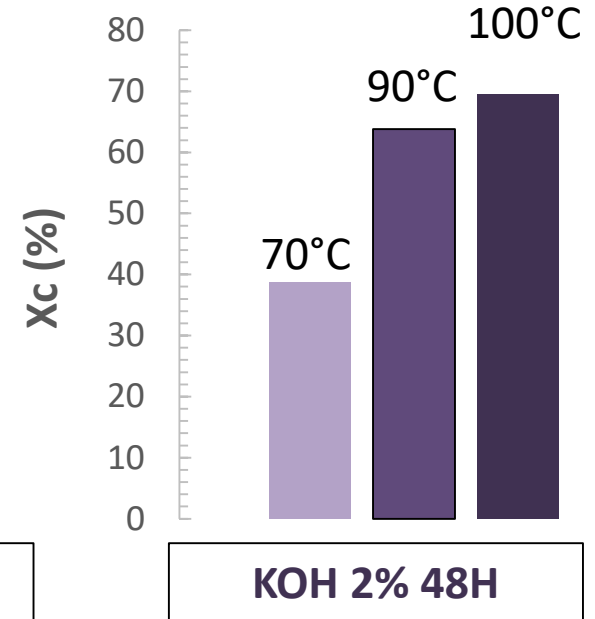
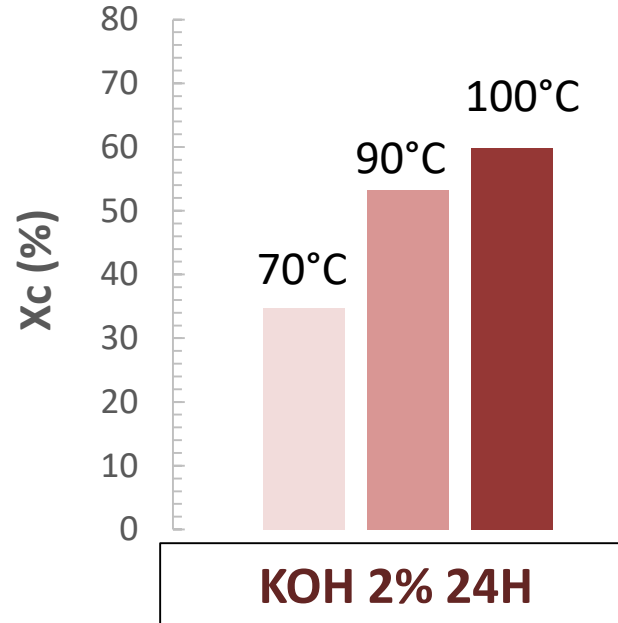
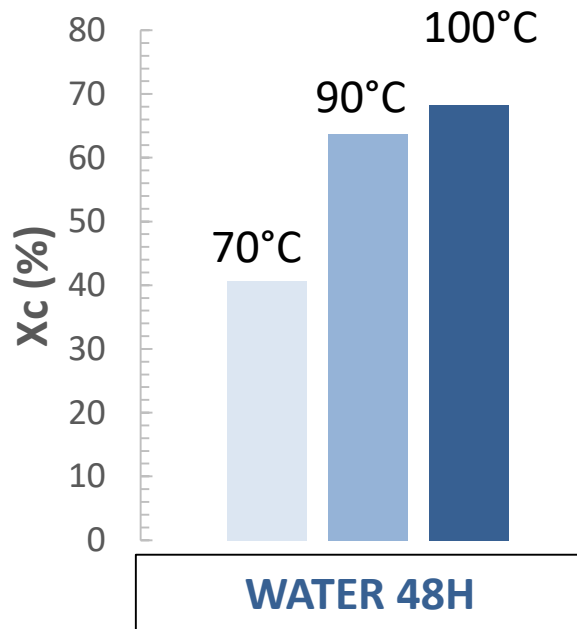
DSC

Semi-crystalline bottle



Semi-crystalline bottle

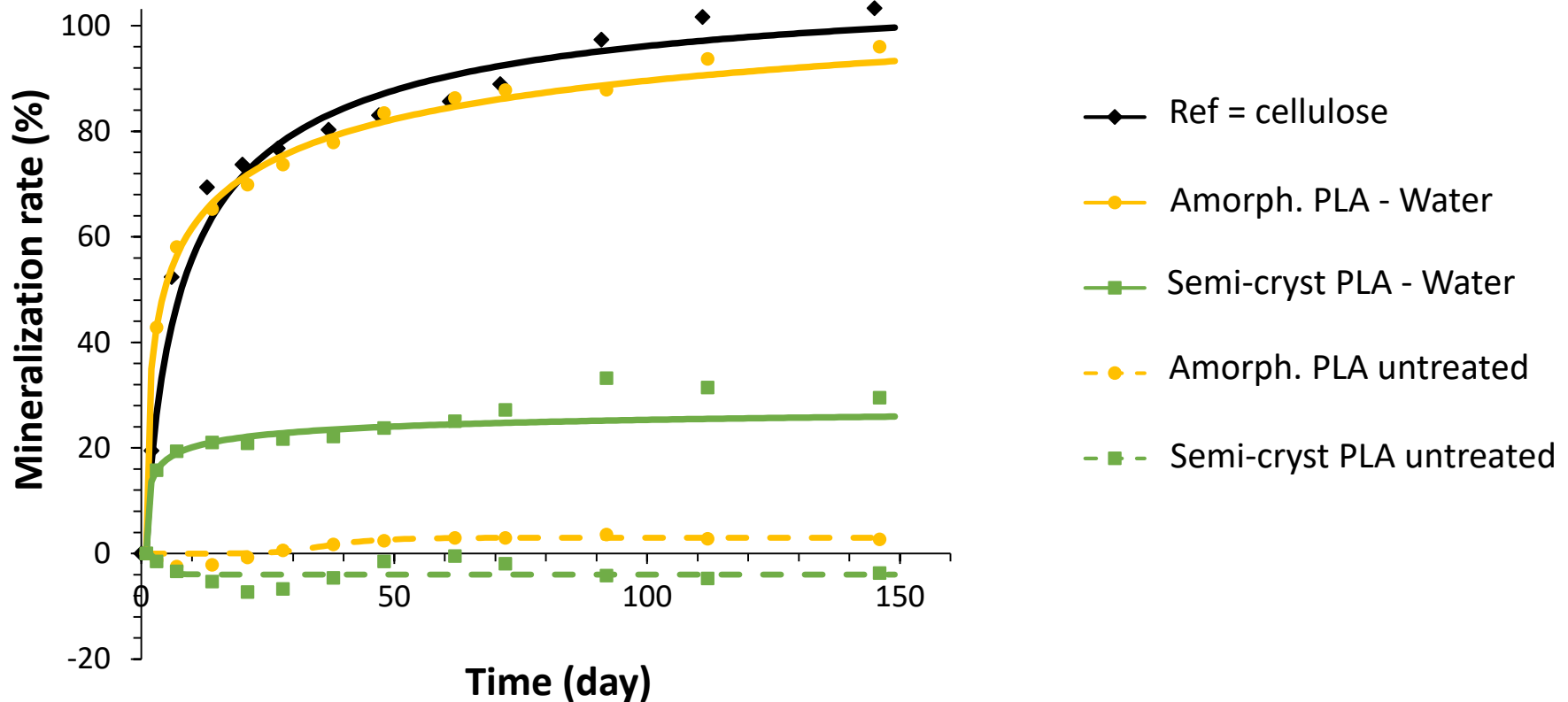
↑ X_c



✓ ↗ Hydrolysis rate => => ↗ crystallinity rate

Home composting conditions (28°C)

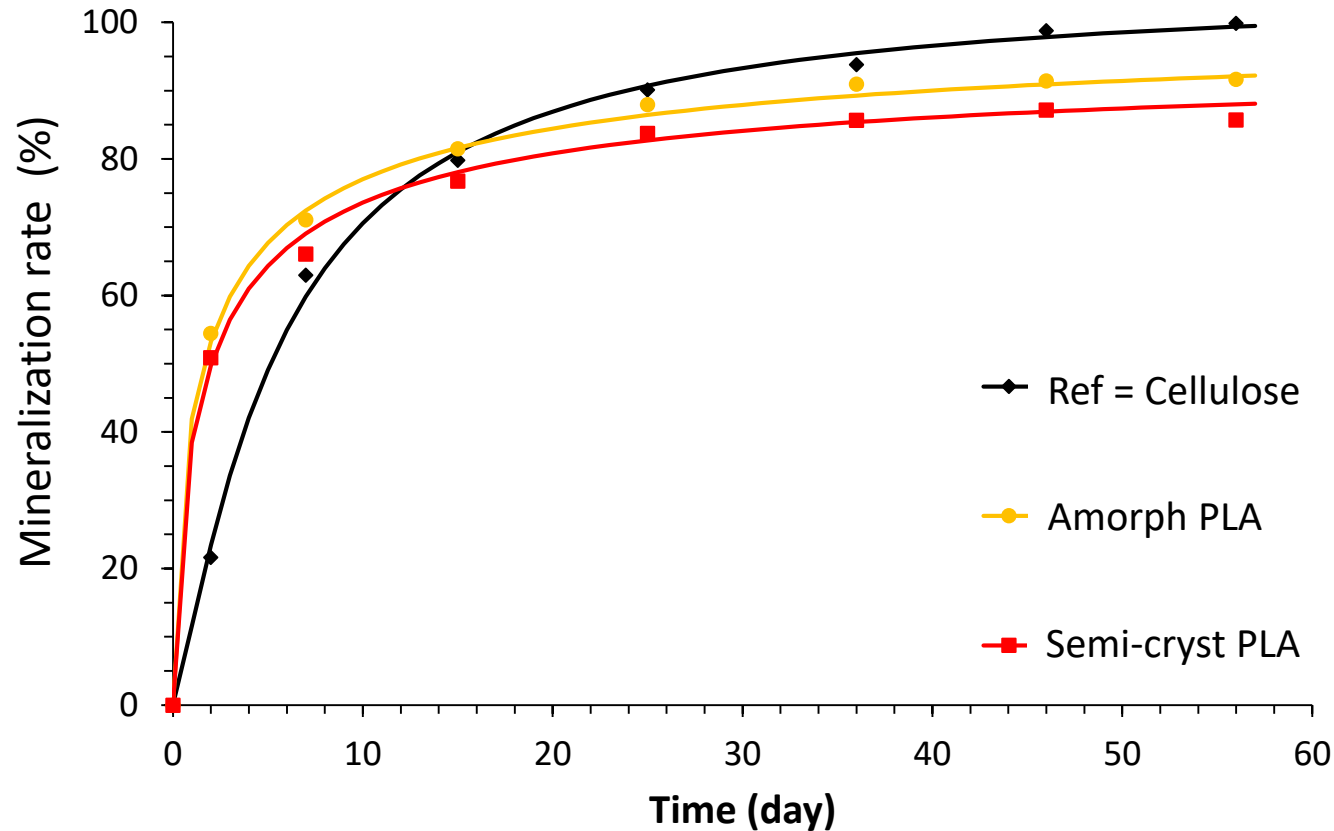
Water treatment @ 100°C



✓ Biodegradation in mesophilic composting conditions improved but not achieved for semicrystalline PLA after thermal water treatment

Home composting conditions (28°C)

KOH 2% treatment @ 70°C



✓ Biodegradation in mesophilic composting conditions achieved for both amorphous and semicrystalline PLA after thermal KOH treatment

Among the different methodological approaches used to monitor the degradation status of polymer during the biodegradation process

- the changes in thermal properties and notably **the decrease in T_g** and **the increase in X_c** appear as relevant indicators of molar mass decrease

Perspectives

- Study the end-of-life performance of biodegradable plastics and packaging through the existing organic valorization routes
- Improve the biodegradability of PLA and PBAT in mesophilic conditions
- Provide data demonstrating the non-persistence of residual compostable micro-fragments



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Many thanks for attention

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